



Deuxième école RMN du  
GERM  
Cargèse 2008



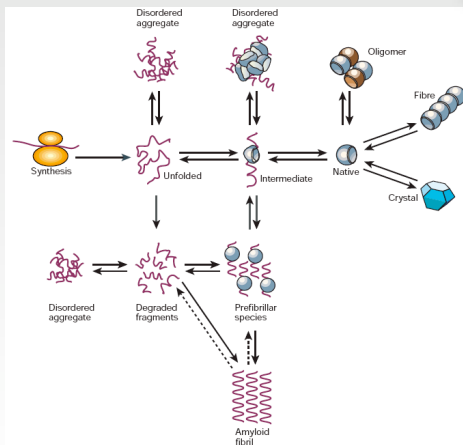
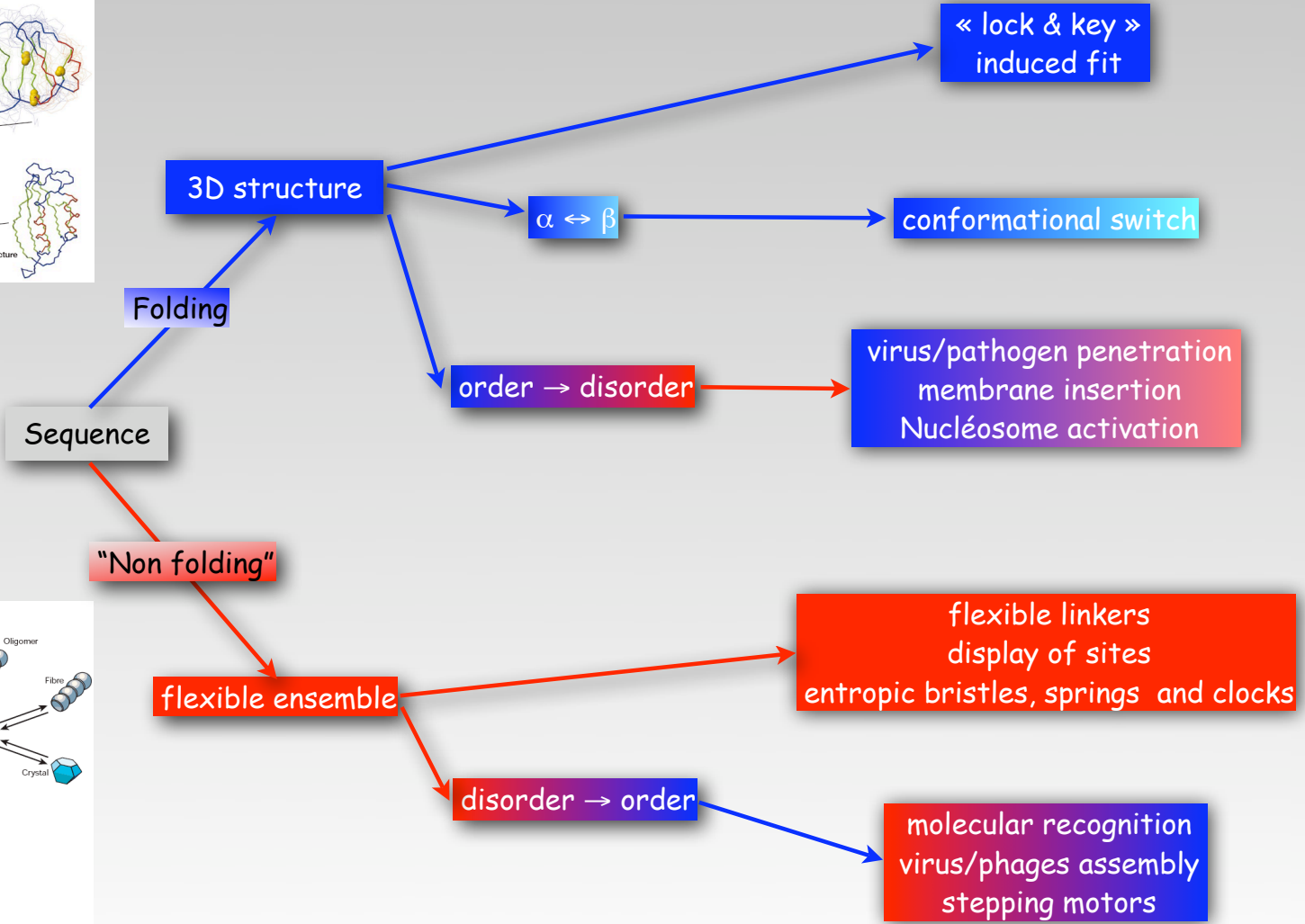
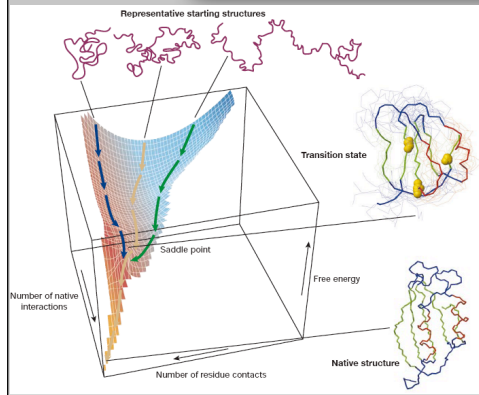
Analysis of micro-millisecond timescale  
motions  
of macromolecules in liquid state  
by Nuclear Magnetic Resonance

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0169823794

# A view of proteins multiple states



# Techniques to monitor protein dynamics

X-ray cristallography	B factors	time scales static disorder, crystal contacts, ...
X-Ray, neutron scattering Doniach, <i>Chem. Rev.</i> 2001, <b>101</b> ; Zacai, <i>science</i> 2000, <b>288</b> .	size/shape modifications timescales (ps-ns) for <sup>1</sup> H positions	
Fluorescence Weiss, <i>Nat. Struct. Biol.</i> 2000, <b>7</b> ; Yang, <i>Science</i> 2003, <b>302</b> ; Haustein, <i>Curr. Opin. Struct. Biol.</i> 2004, <b>14</b> .	ensemble / single molecule cellular context	probes
Mass Spectroscopy (HX MS) radical footprinting Wales, <i>Mass. Spectrom. Rev.</i> 2006, <b>25</b> ; Busenlehner, <i>Arch. Biochem. Biophys.</i> 2005, <b>433</b> . Guan, <i>Trends. Biochem. Sci.</i> 2005, <b>30</b> .	large moldecular assemblies	
Mössbauer, Raman, 2D infrared spectroscopy		
Molecular dynamics		Forcefields ... Short timescales ...
NMR Boehr, <i>Chem. Rev.</i> 2006, <b>106</b> ,3055. Palmer, <i>Chem. Rev.</i> 2004, <b>104</b> , 3623.	<ul style="list-style-type: none"> <li>⇒ 10<sup>-12</sup>↔ 10<sup>5</sup> s</li> <li>⇒ Site-specific information</li> <li>⇒ multiple atomic probes <sup>1</sup>H, <sup>2</sup>H, <sup>15</sup>N, <sup>13</sup>C, <sup>31</sup>P, ...</li> <li>⇒ Simultaneous monitoring of probes</li> <li>⇒ kinetic &amp; thermodynamic profile of dynamic processes</li> </ul>	<ul style="list-style-type: none"> <li>⇒ isotope labeling</li> <li>⇒ quantities</li> <li>⇒ size limitation</li> <li>⇒ complexity of the method ?</li> </ul>

# Bibliography

## Theoretical NMR basis :

- A. Carrington et A. Mc Lachlan Introduction to Magnetic Resonance with applications to chemistry and chemical physics, Harper International , 1967
- C. Slichter Principle of Magnetic Resonance , 3 ème ed, 1990, Springer Verlag
- A. Abragam Principles of Magnetic Resonance, Oxford University Press, 1961
- Ernst, Bodenhausen, Wokaun Principle of Nuclear Magnetic Resonance in one and two dimensions, Oxford Science 1987
- M. Goldman\* Quantum description of high resolution NMR in liquids. Oxford 1990
- D. Canet\* La RMN : concepts, méthodes et applications (2<sup>ème</sup> ed.), UniverSciences, Dunod 2002
- M. H. Levitt\*\*\* Spin dynamics. Basic of Nuclear Magnetic Resonance (686 pages) J.Wiley 2001
- John Cavanagh, Wayne J. Fairbrother, Arthur G., III Palmer\*\*\* Protein NMR Spectroscopy: Principles And Practice (912 pages) Academic Press; Édition : 2nd (novembre 2006)

## Experimental aspects, applications

- J. K. M. Sanders, B. K. Hunter\* Modern NMR Spectroscopy: A Guide for Chemists 2<sup>nd</sup> Edition, Oxford 2003
- D. Shaw Fourier transform NMR spectroscopy Elsevier 1984
- K. Wüthrich\* NMR of proteins and nucleic acids. Wiley interscience 1986

# How motions are visible in NMR ?

**Molecular motions** influence **NMR parameters**

**Time scale of motions** compared to **NMR characteristic timescales**

⇒ **Three characteristic NMR timescales**

Return of the spin system to equilibrium :  $T_1$  / NMR signal lifetime :  $T_2$

NMR experiment  $\leftrightarrow$  spins system perturbation

Liquid state:  $T_1 \sim 100\text{ms} \leftrightarrow \text{s}$  ;  $T_2 \sim 10\text{ms} \leftrightarrow \text{s}$

⇒ Minimal frequencies of motions that can be characterized during a single NMR experiment.

**Spectral timescale** :  $\tau = 1/\Delta\nu$

spectral features : chemical shifts range, couplings, ... Hz  $\leftrightarrow$  kHz

⇒ Averaging of interactions by motions faster than the spectral dispersion due to these interactions.

⇒ Perturbation of spectral features by motions in the same range than the spectral timescale.

**Larmor timescale** :  $\omega_0$

precession frequency of spins in a magnetic field ( $B_0 = \omega_0/\gamma$ )

⇒ transitions efficiency between spin states is determined by molecular fluctuations (spectral densities fo motions) at these frequencies.

# How motions are visible in NMR ?

Molecular motions influence NMR parameters

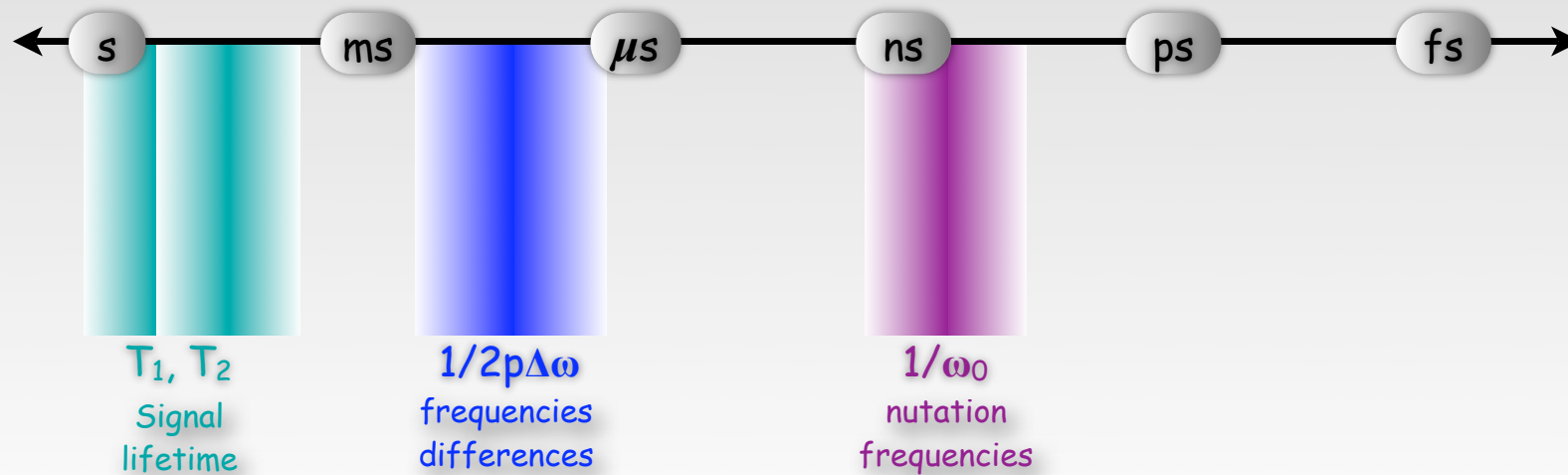
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Spectral timescale :  $\tau = 1/\Delta\nu$

Larmor timescale :  $\omega_0$



# Motions & function & NMR

Functions

Folding ; order ↔ disorder

Interactions

catalytic processes  
Regulation, signalization

Internal motions

Types of motions

Macroscopic diffusion

Conformational exchange

Molecular rotations

Molecular vibrations

Effect of motions on spins interaction

Diffusion experiments

Magnetization exchange

Lineshape modifications

Spin relaxation

Secular interactions averaging

Averaging of spectral components

NMR parameters

$T_1, T_2$   
Signal lifetime

$1/\Delta\omega$   
frequencies differences

$1/\omega_0$   
nutations frequencies



# NMR timescales -1- Longitudinal relaxation time $T_1$

✓  $T_1$  characterizes the time for a spin system to reach equilibrium.

✓ Defines interscan delay

✓ Motions slower than  $T_1$  cannot be characterized by a single spectra.

✓ NB.  $T_1$  depends on magnetic field  $B_0$ , nature of spins, molecule size, local flexibility, temperature, etc.

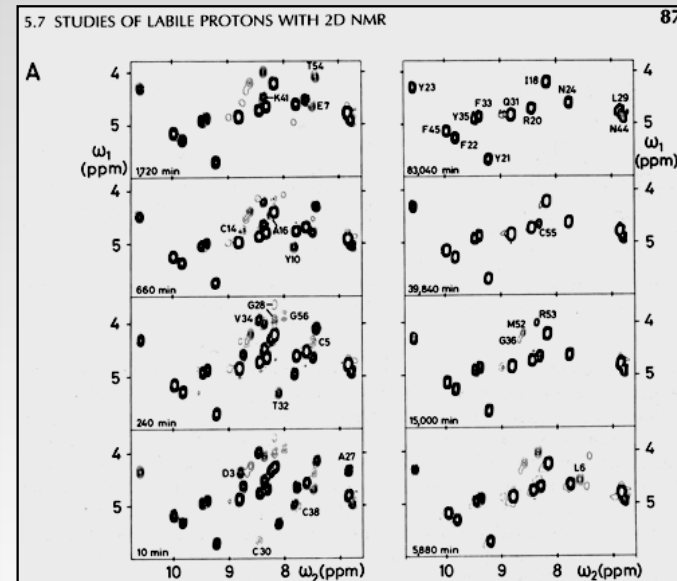
## Dynamical processes slower than $T_1$

Real time folding

Transient / out of equilibrium experiments

Protein - ligand interactions

H-N  $\leftrightarrow$  D-N exchange rate constants



Wüthrich, « NMR of proteins and nucleic acids »,  
Wiley Interscience, 1986



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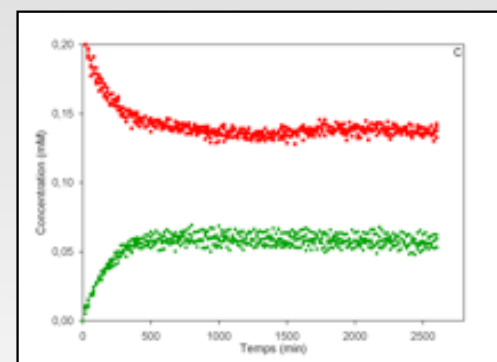
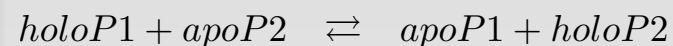
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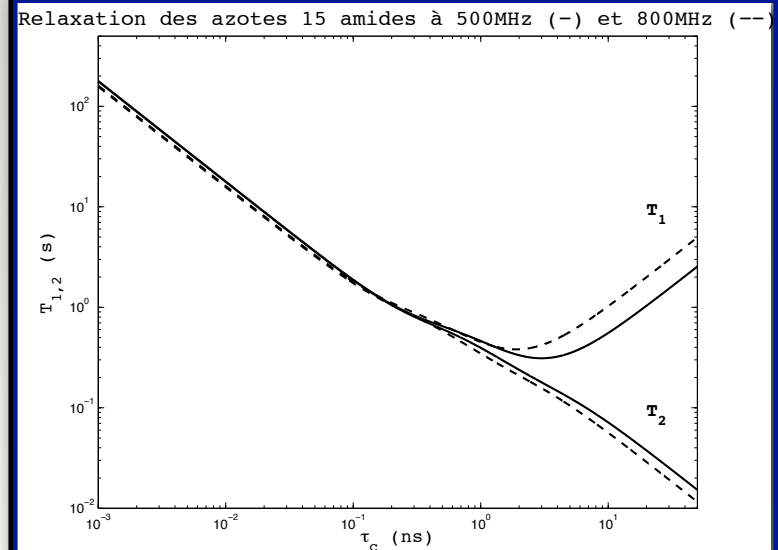
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magnetic field, molecular size and correlation time dependence of  $T_1$  and  $T_2$



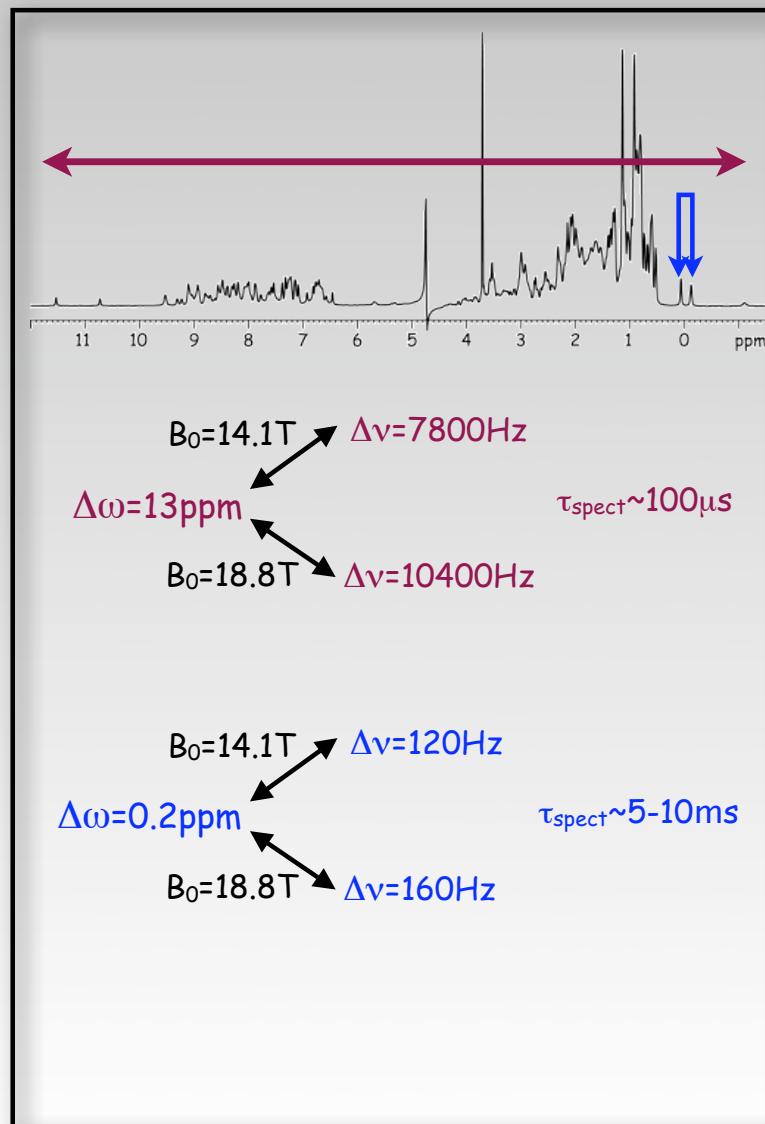
# NMR timescales -2- Spectral or "chemical shift" timescale

- ✓ Defined by the observed spectral width
- ✓ More precisely by the difference of resonance frequency between two spins from two different nuclei or from the same nuclei in two different states.

$$\Delta\nu(\text{Hz}) = \frac{\Delta\omega(\text{ppm})}{2\pi} * 10^{-6} * \gamma B_0$$

$$\tau_{\text{spect}} = \frac{1}{\Delta\nu(\text{Hz})}$$

- ✓ Motions slower than  $\Delta\nu$  have no effect on spectral feature
- ✓  $\Delta\nu$  depends on magnetic field and on spins nature



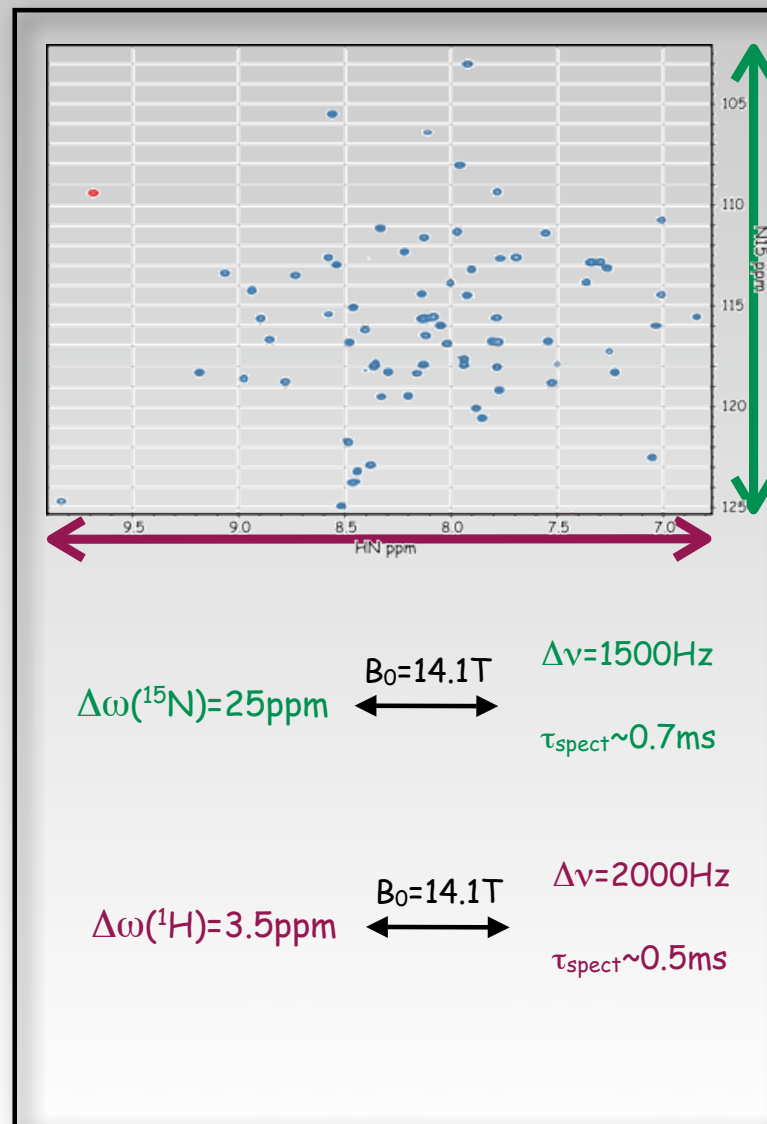
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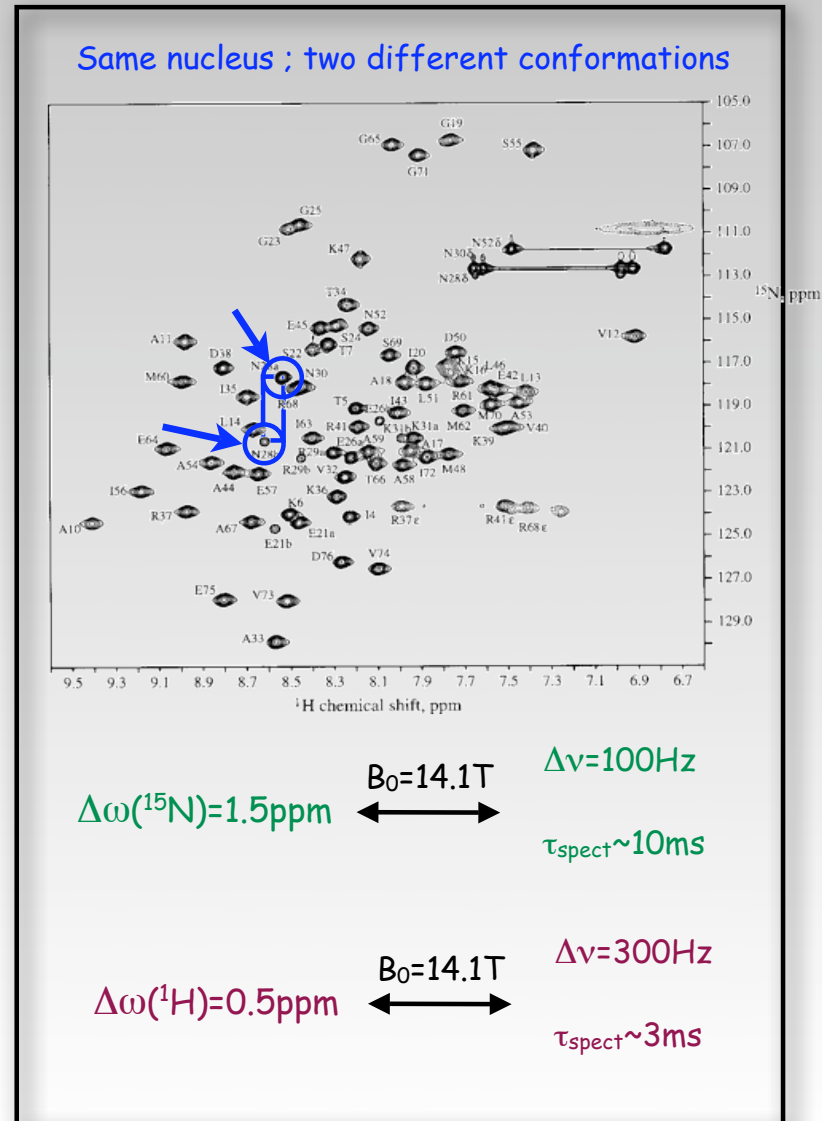
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### Averaging of secular interactions by motions

✓ Dipolar interaction between spins magnetic moments.

$$\hat{H}_{IS}^{\text{DD}} = -\frac{\mu_0\gamma_I\gamma_S\hbar}{4\pi r_{IS}^3} \left( \frac{3\cos^2\theta_{IS} - 1}{2} \right) (3\hat{I}_z\hat{S}_z - \hat{\mathbf{I}}\hat{\mathbf{S}})$$

$$E/\hbar < 10^4 - 10^5 \text{ Hz}$$

✓ Chemical shift anisotropy

$$\hat{H}_I^{\text{CSA}} = -\gamma_S \frac{c_{\parallel} - c_{\perp}}{3} B_0 (3\cos^2\theta - 1) \hat{I}_z$$

$$E/\hbar < 10^4 - 10^5 \text{ Hz}$$

✓ Interaction spin quadrupole/electric field

$$\hat{H}_I^{\text{Q}} \equiv \omega_I^{\text{Q}} (3\hat{I}_z^2 - \hat{\mathbf{I}}\hat{\mathbf{I}}); \omega_I^{\text{Q}} = \frac{3eQ_I}{4I(2I-1)} V_{zz}^I(\theta)$$

$$I > \frac{1}{2} \quad E/\hbar \approx 2 \cdot 10^5 \text{ Hz} \left( {}^2\text{H} \right) \text{ and } 3 \cdot 10^6 \left( {}^{14}\text{N} \right)$$

✓ Unpaired electron

## NMR timescales -2- Spectral or "chemical shift" timescale

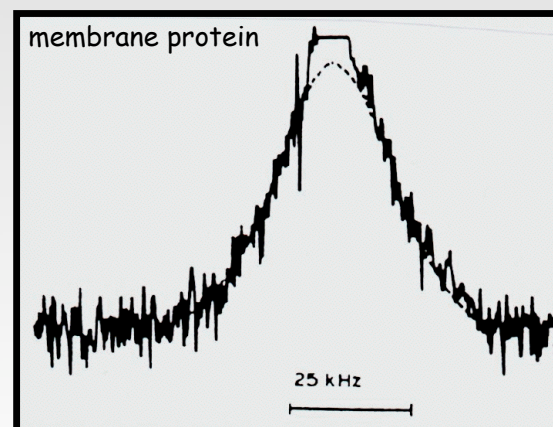
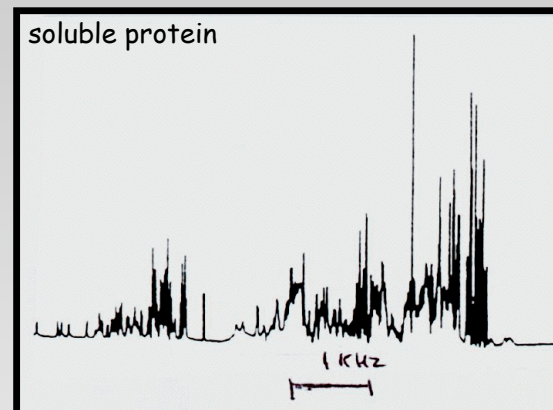
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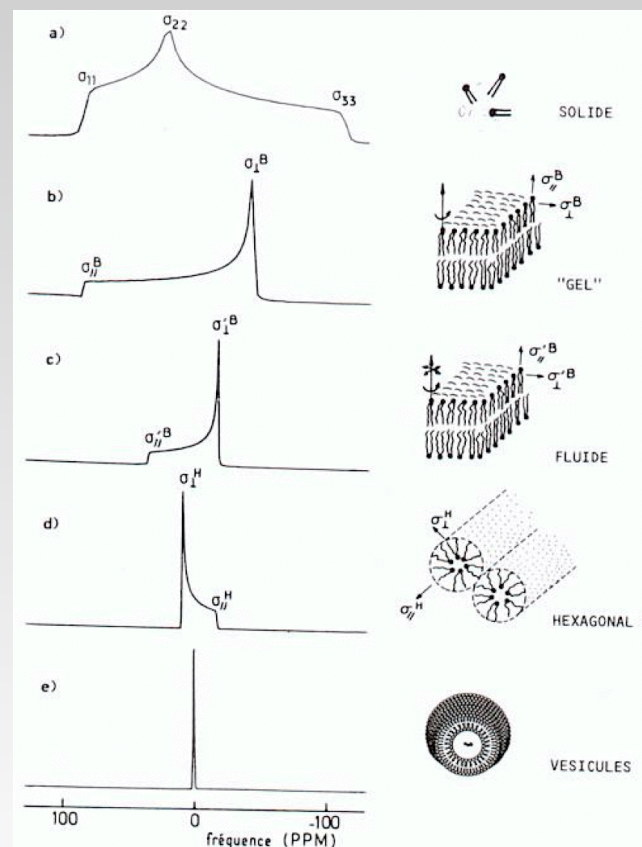
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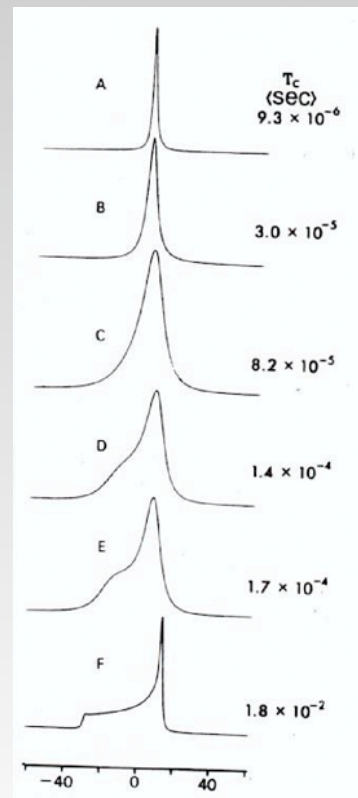
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### Averaging of secular interactions by motions



Averaging of chemical shift anisotropy ( $^{31}\text{P}$ )

Burnell et al., Biochim. Biophys. Acta 603, 63 (1980)

# NMR timescales -2- Spectral or "chemical shift" timescale

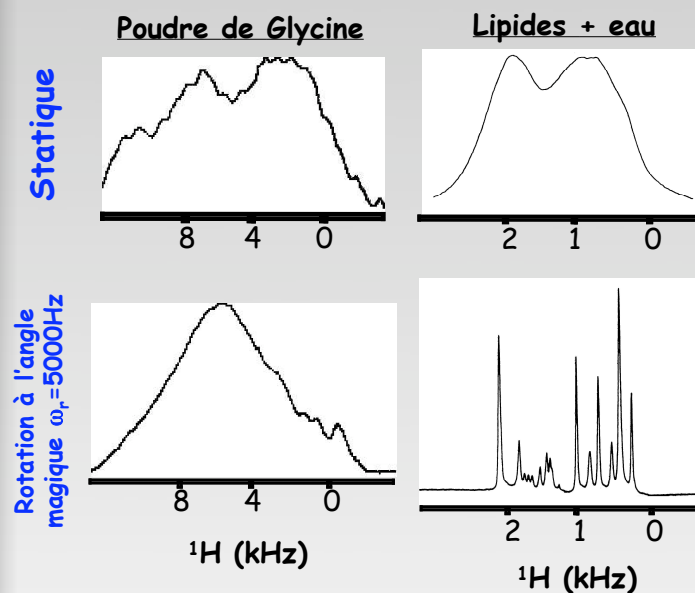
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motions generated to average interactions



Davis, Auger & Hodges *Biophysical Journal* **69**:1917-1932 (1995)  
 Gross et al., *J. Magn. Res.* **106**, 187-190 (1995)  
 Carlotti, Aussenac & Dufourc *Biochim. Biophys. Acta.* **1564**:156-164 (2002)

## NMR timescales -3- The "Larmor" timescale

✓ Defined by the resonance frequencies of spins

✓ i.e. the energy difference between spins states levels

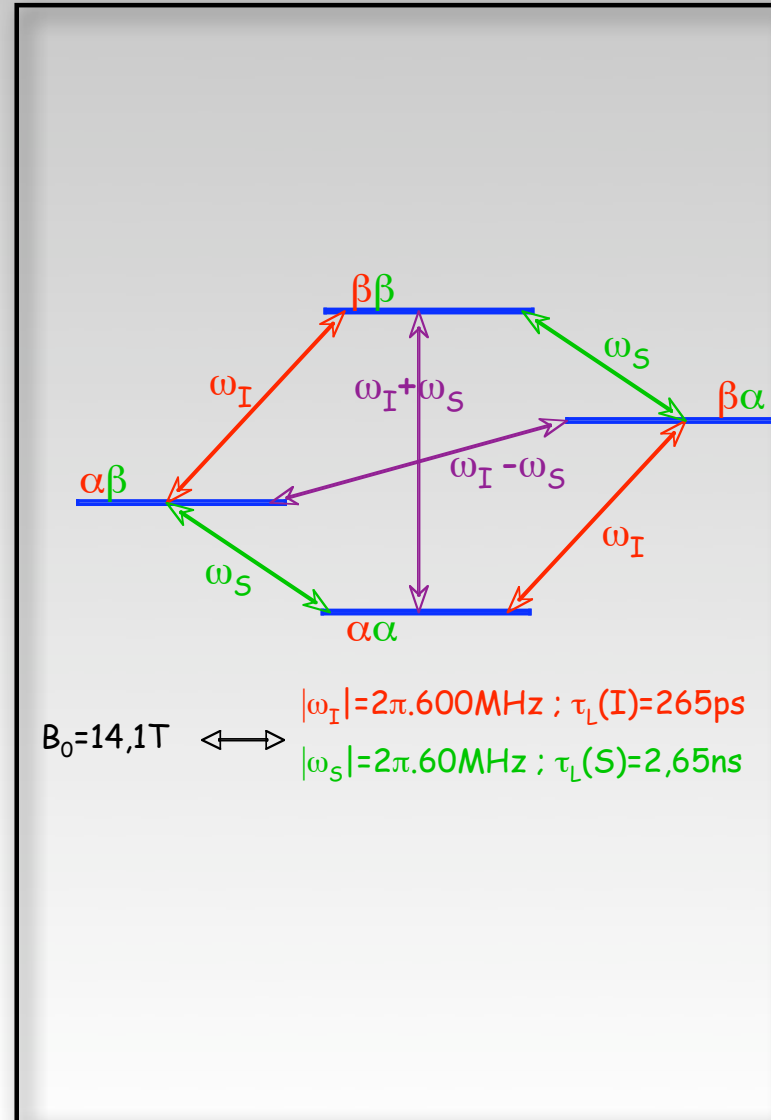
$$\omega_0 = -\gamma B_0$$

$$\tau_{Larmor} = \frac{1}{|\omega_0|} = \frac{1}{2\pi\nu_0}$$

✓ Motions in these timescale have no direct effect on spectra.

✓ Motions in these timescales are responsible for the efficiency of spins relaxation processes

✓ The relationship between motions and relaxation rate/time constants is not simple



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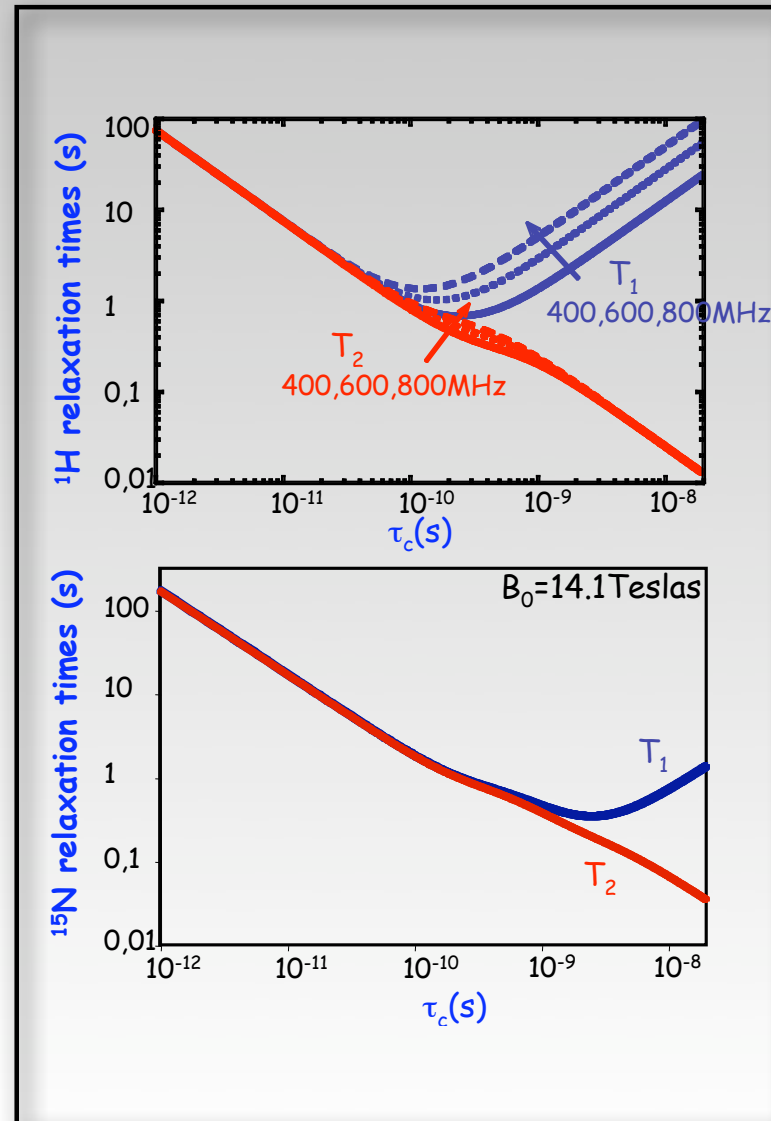
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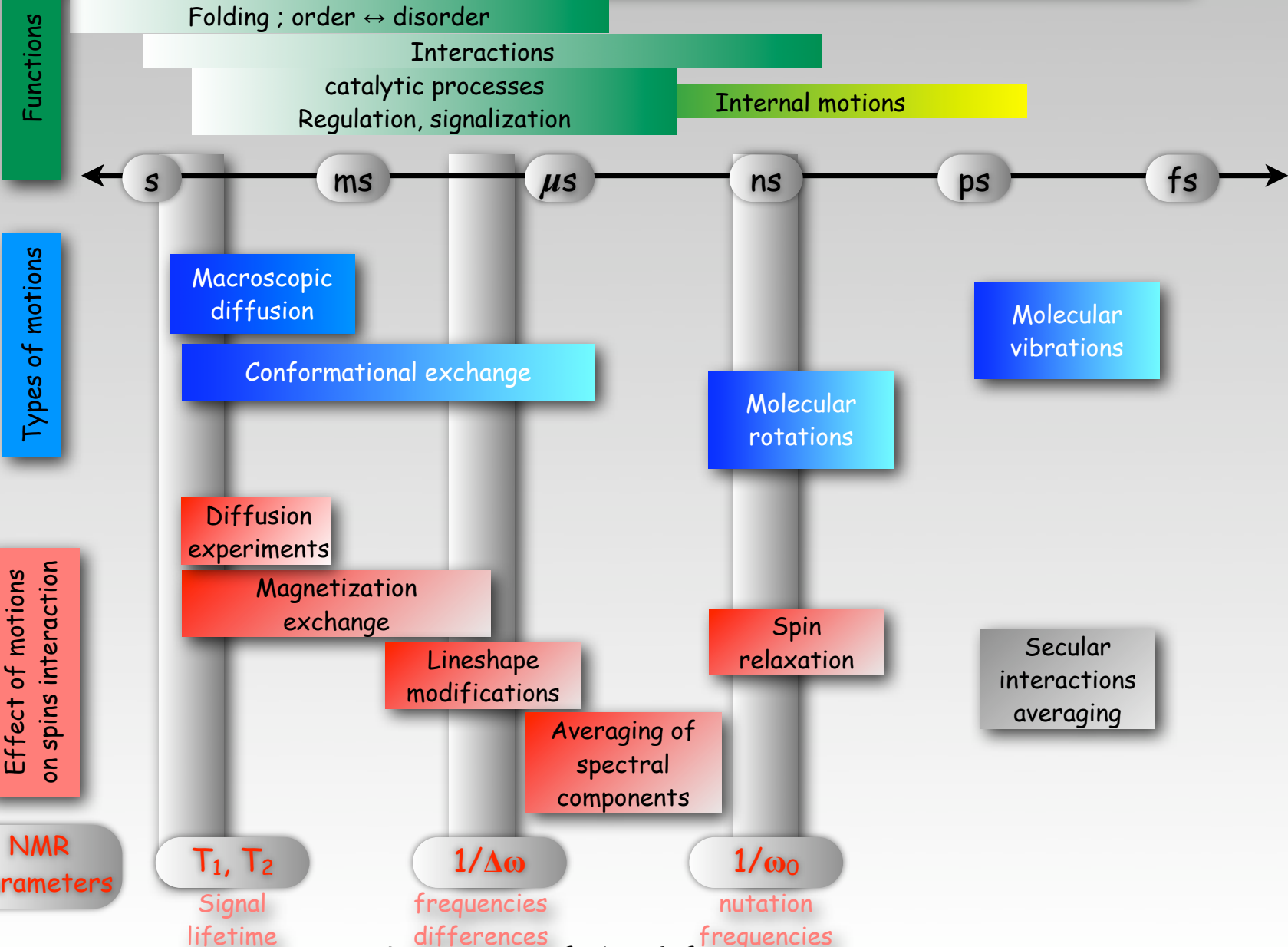
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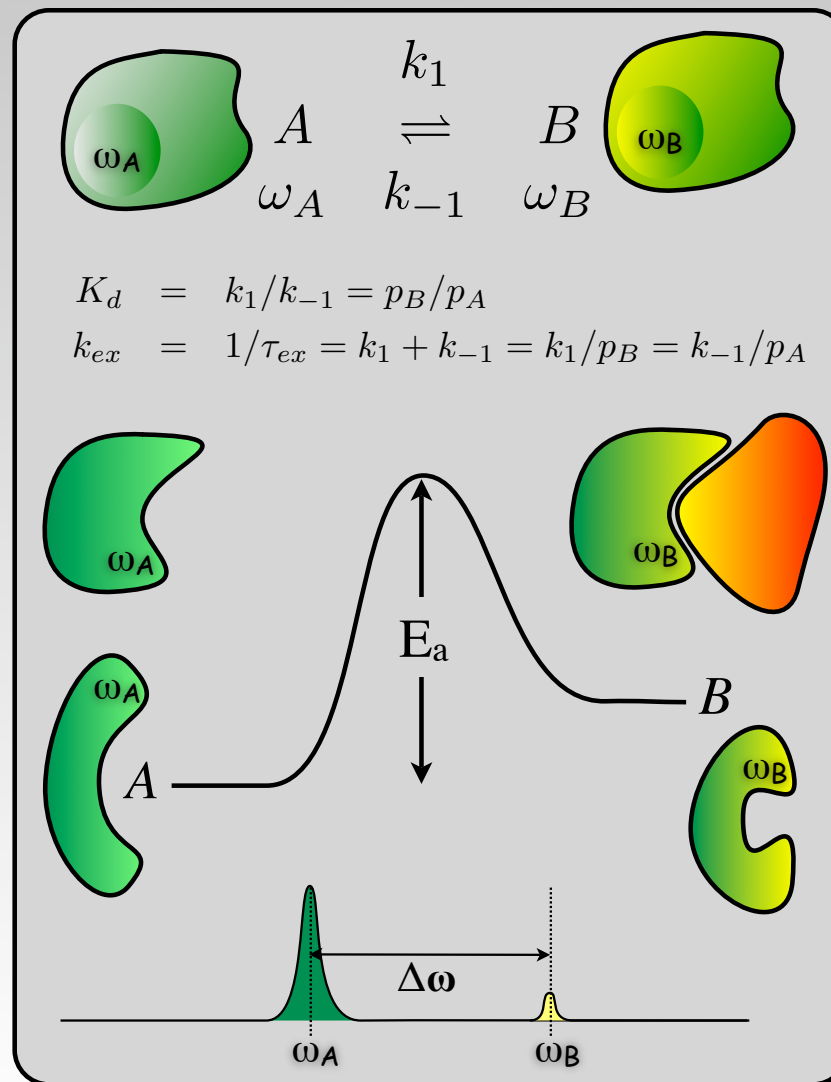
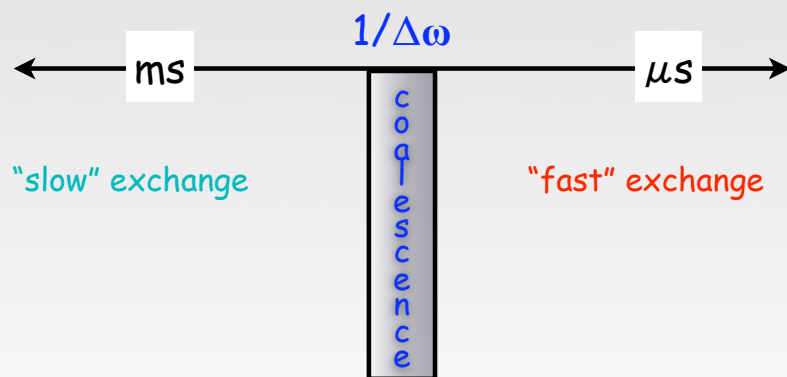
# Motions & function & NMR



# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

## Conformational exchange

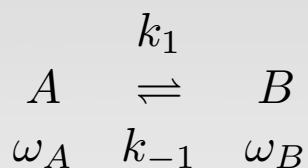
- ✓ Motions in the  $\mu\text{s}$ - $\text{ms}$  range induce spectral modifications.
- ✓ These motions usually correspond to conformational exchange conformationnel, or chemical exchange.
- ✓ Their effect on spectra depends of relative values of  $k_{ex}$  ( $\tau_{ex}$ ) and  $\Delta\omega$  ( $1/\Delta\omega$ ). They are called either **slow**, **intermediate** or **fast** exchange processes at the chemical shift timescale.



# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

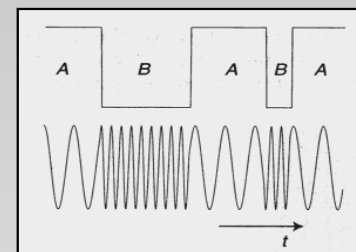
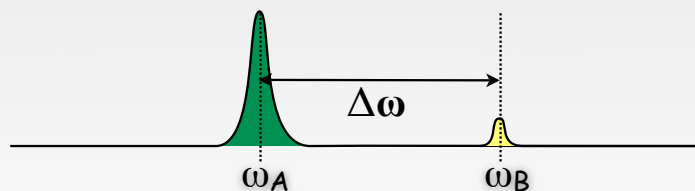
## Conformational exchange

- ✓ Conformational exchange is characterized by a probability  $k_{\text{ex}}$  to switch from one state to the other, associated with an instantaneous change of precession frequency.
- ✓ These sudden frequency changes induce a dephasing of the transverse magnetization, which is added to the natural loss of coherence of spins.

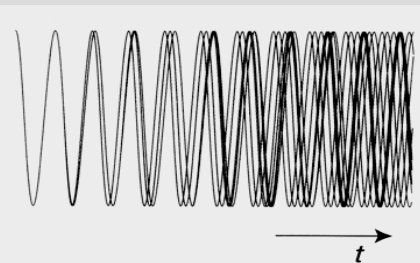


$$K_d = k_1/k_{-1} = p_B/p_A$$

$$k_{\text{ex}} = 1/\tau_{\text{ex}} = k_1 + k_{-1} = k_1/p_B = k_{-1}/p_A$$

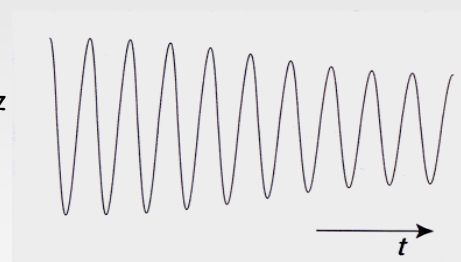


20 molecules  
in state A



$\Delta\nu=1000\text{Hz}$   
 $k_{\text{ex}}=500\text{Hz}$   
 $p_A=p_B$

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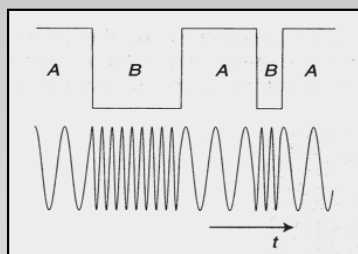


Resulting transverse magnetization for a great number of spins initially in the same state (effect of exchange only)

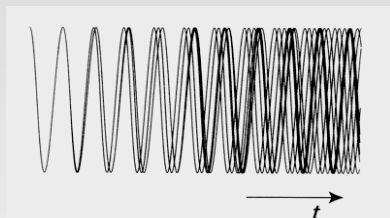
# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

## Conformational exchange

⇒ Modification of spectral features

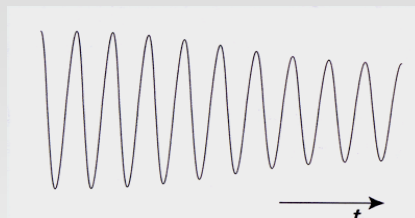


⇒ depends of the relative values of  $k_{ex}$  and  $\Delta\omega$



20 molecules  
in state A

$$\begin{aligned} \Delta\omega/2\pi &= 1000\text{Hz} \\ k_{ex} &= 500\text{Hz} \\ p_A &= p_B \end{aligned}$$

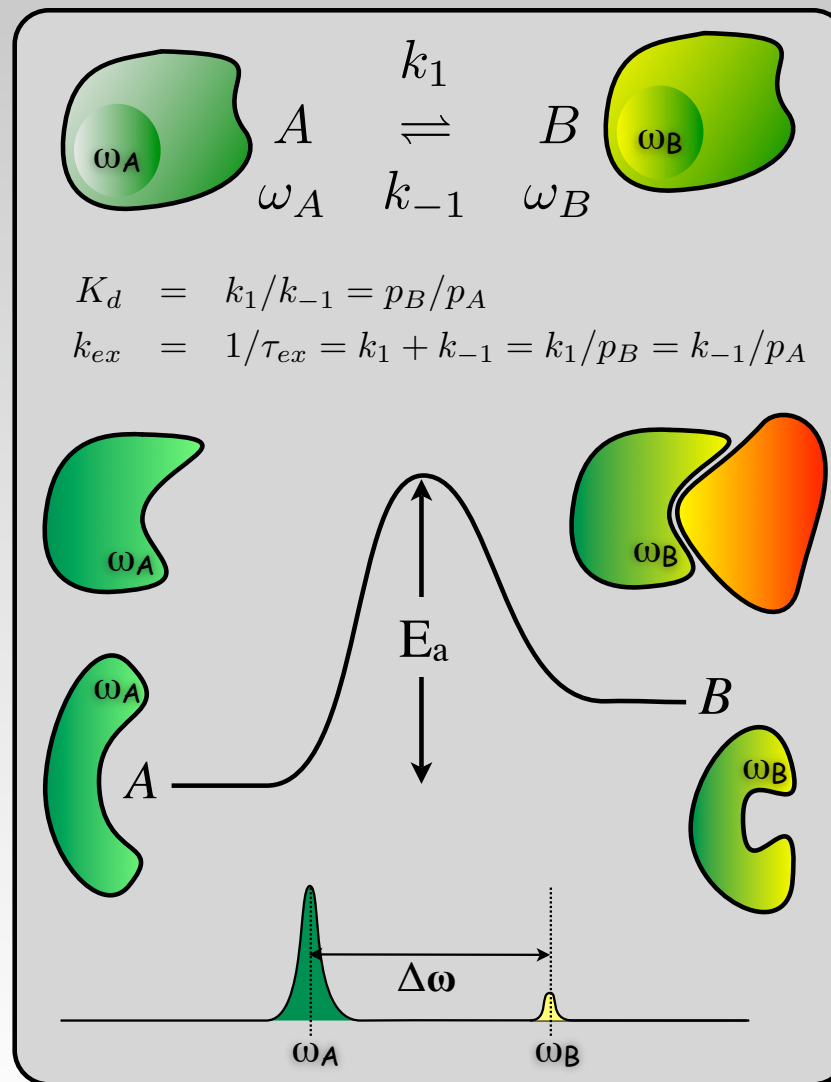


⇒ Additional dephasing of the magnetization

⇒ Apparent enhanced transverse relaxation

$$\Rightarrow R_2^{app} = R_2 + R_{ex}$$

⇒ Larger linewidths

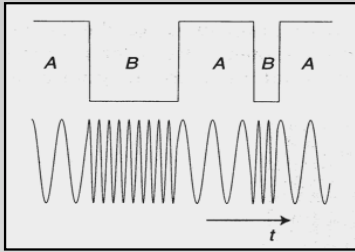




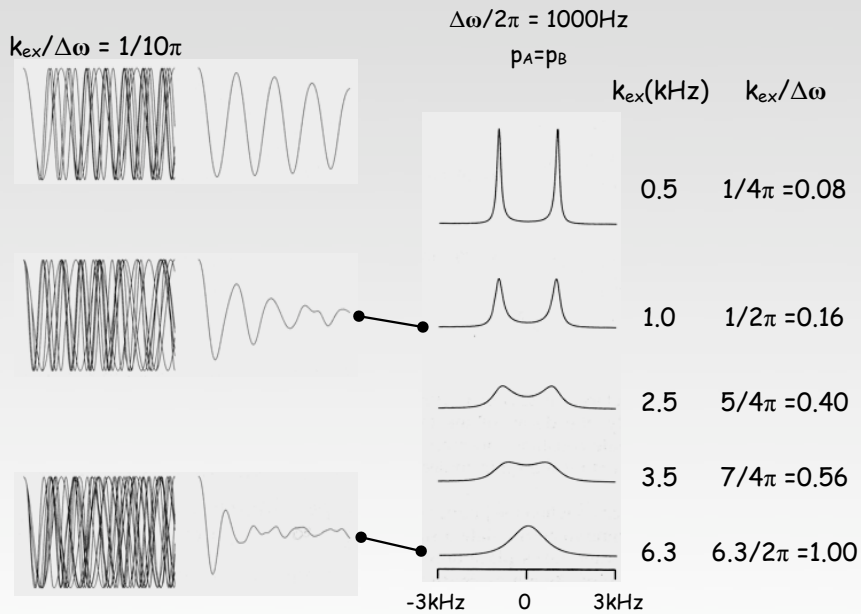
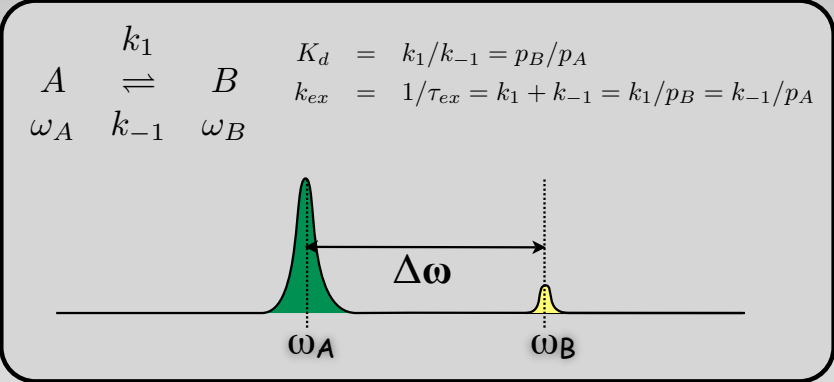
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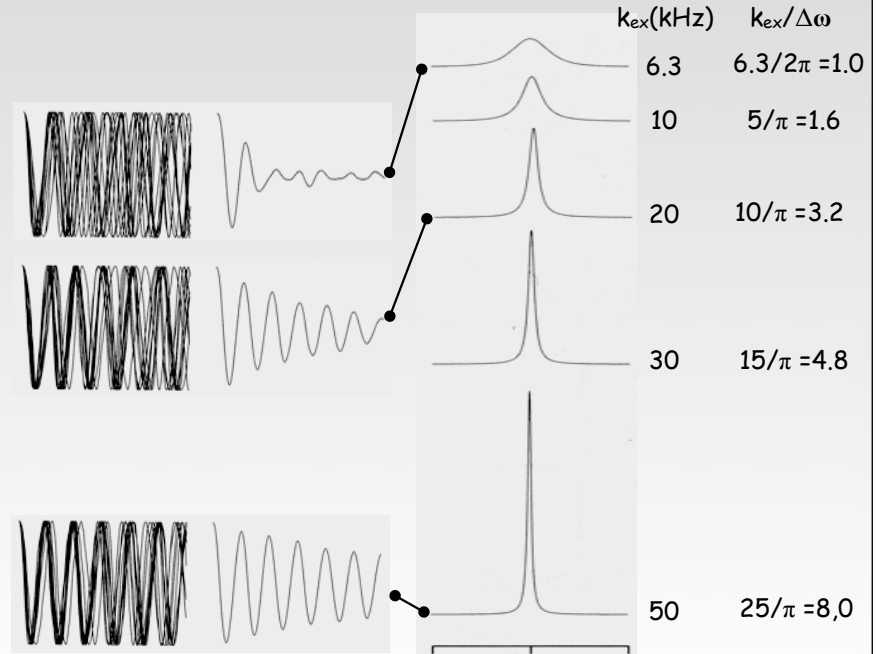
⇒ Modification of spectral features



⇒ depends of the relative values of  $k_{ex}$  and  $\Delta\omega$



Motional broadening



Motional narrowing

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## Conformational exchange

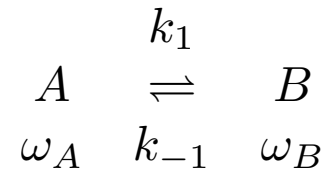
General equation for a system undergoing conformational exchange

$$\bar{M}(t) = \begin{bmatrix} \bar{M}_1(t) \\ \bar{M}_2(t) \\ \dots \\ \bar{M}_n(t) \end{bmatrix}$$

$$\frac{d}{dt} \Delta \bar{M}_z(t) = (-R + K) \Delta \bar{M}_z(t)$$

$$\frac{d}{dt} \bar{M}_+(t) = (i\Omega - R + K) \bar{M}_+(t)$$

The case of a two state exchange



$$R - K = \begin{bmatrix} \rho_A - k_1 & -k_{-1} \\ -k_1 & \rho_B - k_{-1} \end{bmatrix}$$

$$\frac{d}{dt} \begin{bmatrix} \Delta M_A^+(t) \\ \Delta M_B^+(t) \end{bmatrix} = \begin{bmatrix} -i\Omega_A - R_{2A}^0 - \rho_B k_{ex} & \rho_A k_{ex} \\ \rho_B k_{ex} & -i\Omega_B - R_{2B}^0 - \rho_A k_{ex} \end{bmatrix} \begin{bmatrix} M_A^+(t) \\ M_B^+(t) \end{bmatrix}$$

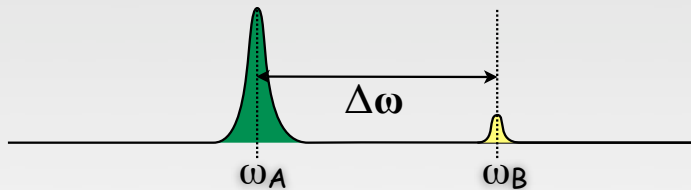
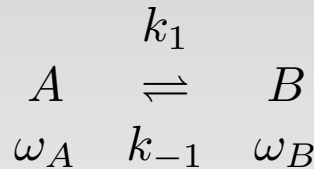
$$\begin{bmatrix} M_A^+(t) \\ M_B^+(t) \end{bmatrix} = \begin{bmatrix} a_{AA}(t) & a_{AB}(t) \\ a_{BA}(t) & a_{BB}(t) \end{bmatrix} \begin{bmatrix} M_A^+(0) \\ M_B^+(0) \end{bmatrix}$$

In a general case, the expression of the matrix coefficients is complicated

# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

## Conformational exchange

General equation  
for a two state exchange



$$\begin{bmatrix} M_A^+(t) \\ M_B^+(t) \end{bmatrix} = \begin{bmatrix} a_{AA}(t) & a_{AB}(t) \\ a_{BA}(t) & a_{BB}(t) \end{bmatrix} \begin{bmatrix} M_A^+(0) \\ M_B^+(0) \end{bmatrix}$$

$$a_{AA}(t) = \frac{1}{2} \left[ \left( 1 - \frac{-i\omega_A + i\omega_B + \rho_A - \rho_B + k_1 - k_{-1}}{(\lambda_+ - \lambda_-)} \right) \exp(-\lambda_- t) + \left( 1 + \frac{-i\omega_A + i\omega_B + \rho_A - \rho_B + k_1 - k_{-1}}{(\lambda_+ - \lambda_-)} \right) \exp(-\lambda_+ t) \right]$$

$$a_{BB}(t) = \frac{1}{2} \left[ \left( 1 + \frac{-i\omega_A + i\omega_B + \rho_A - \rho_B + k_1 - k_{-1}}{(\lambda_+ - \lambda_-)} \right) \exp(-\lambda_- t) + \left( 1 - \frac{-i\omega_A + i\omega_B + \rho_A - \rho_B + k_1 - k_{-1}}{(\lambda_+ - \lambda_-)} \right) \exp(-\lambda_+ t) \right]$$

$$a_{AB}(t) = \frac{k_{-1}}{\lambda_+ - \lambda_-} \left[ \exp(-\lambda_- t) - \exp(-\lambda_+ t) \right]$$

$$a_{BA}(t) = \frac{k_1}{\lambda_+ - \lambda_-} \left[ \exp(-\lambda_- t) - \exp(-\lambda_+ t) \right]$$

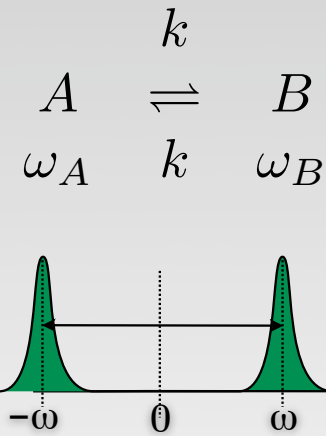
$$\lambda_{\pm} = \frac{1}{2} \left\{ -i\omega_A - i\omega_B + \rho_A + \rho_B + k_1 + k_{-1} \pm \left[ (-i\omega_A + i\omega_B + \rho_A - \rho_B + k_1 - k_{-1})^2 + 4k_1k_{-1} \right]^{1/2} \right\}$$

# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

## Conformational exchange

An easier case

The two state symmetrical exchange



$$\begin{bmatrix} M_A^+(t) \\ M_B^+(t) \end{bmatrix} = \begin{bmatrix} a_{AA}(t) & a_{AB}(t) \\ a_{BA}(t) & a_{BB}(t) \end{bmatrix} \begin{bmatrix} M_A^+(0) \\ M_B^+(0) \end{bmatrix}$$

$$\begin{aligned}
 a_{AA}(t) &= \frac{1}{2} \left[ \left( 1 + \frac{i\omega}{\Delta} \right) \exp\{-(\rho + k - \Delta)t\} + \left( 1 - \frac{i\omega}{\Delta} \right) \exp\{-(\rho + k + \Delta)t\} \right] \\
 a_{BB}(t) &= \frac{1}{2} \left[ \left( 1 - \frac{i\omega}{\Delta} \right) \exp\{-(\rho + k - \Delta)t\} + \left( 1 + \frac{i\omega}{\Delta} \right) \exp\{-(\rho + k + \Delta)t\} \right] \\
 a_{AB}(t) &= a_{BA}(t) = \frac{k}{2\Delta} \left[ \exp\{-(\rho + k - \Delta)t\} - \exp\{-(\rho + k + \Delta)t\} \right] \\
 \Delta &= (k^2 - \omega^2)^{1/2}
 \end{aligned}$$

**Fast exchange**  $k_{ex} \gg \Delta\omega$

$$a_{AA}(t) = a_{BB}(t) = \frac{1}{2} [1 + \exp(-2kt)] \exp(-\rho t)$$

$$a_{AB}(t) = a_{BA}(t) = \frac{1}{2} [1 - \exp(-2kt)] \exp(-\rho t)$$

$$M^+(t) = M_A^+(t) + M_B^+(t) = M^+(0) \exp(-\rho t)$$

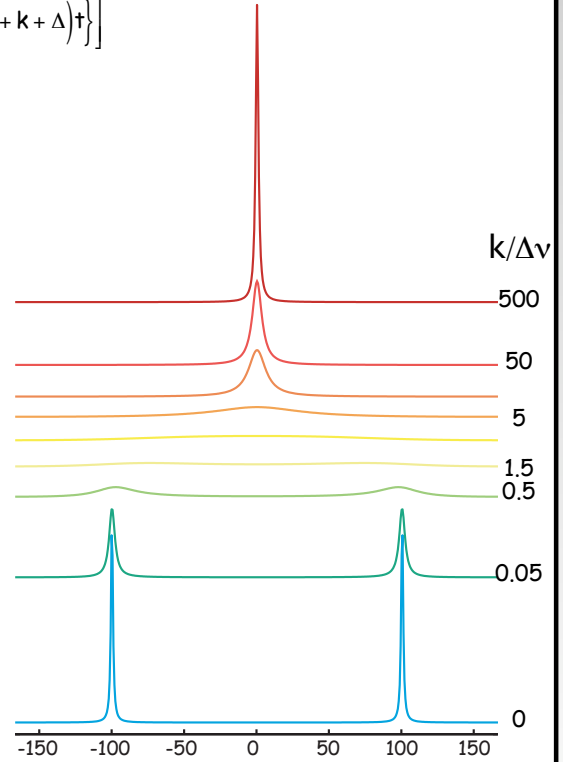
**Coalescence**  $k_{ex} \sim \Delta\omega$

**Slow exchange**  $k_{ex} \ll \Delta\omega$

$$a_{AA}(t) = \exp\{-(\rho + k - i\omega)t\}$$

$$a_{BB}(t) = \exp\{-(\rho + k + i\omega)t\}$$

$$a_{AB}(t) = a_{BA}(t) = 0$$



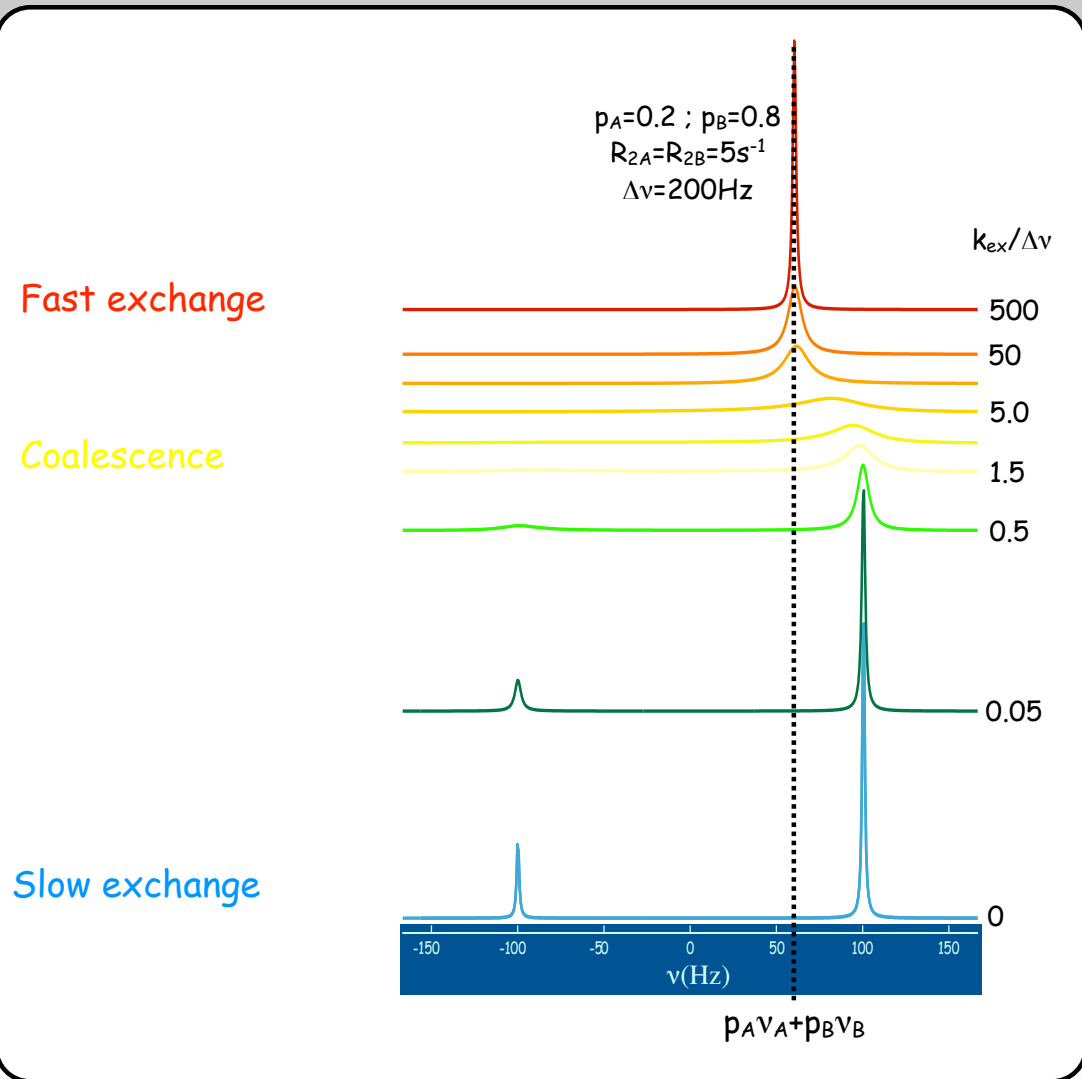
# Analysis of $\mu\text{s}$ -ms motions by NMR

## Conformational exchange

two state non symmetrical exchange

The minor component is more broadened than the major one

When the population difference is large, the minor component can become invisible, even in a "slow" exchange case.

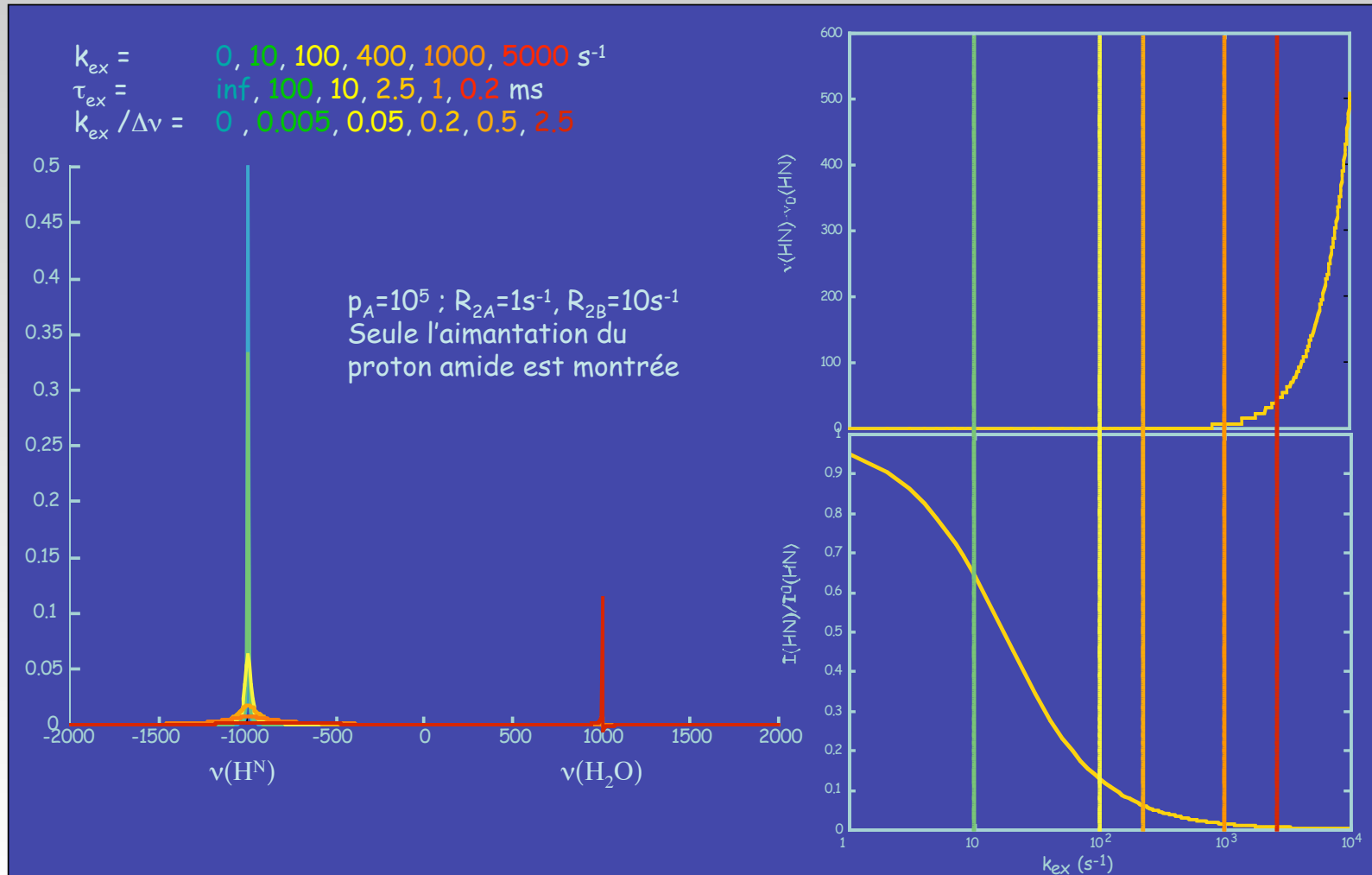


# Analysis of $\mu\text{s}$ -ms motions by NMR

## Conformational exchange

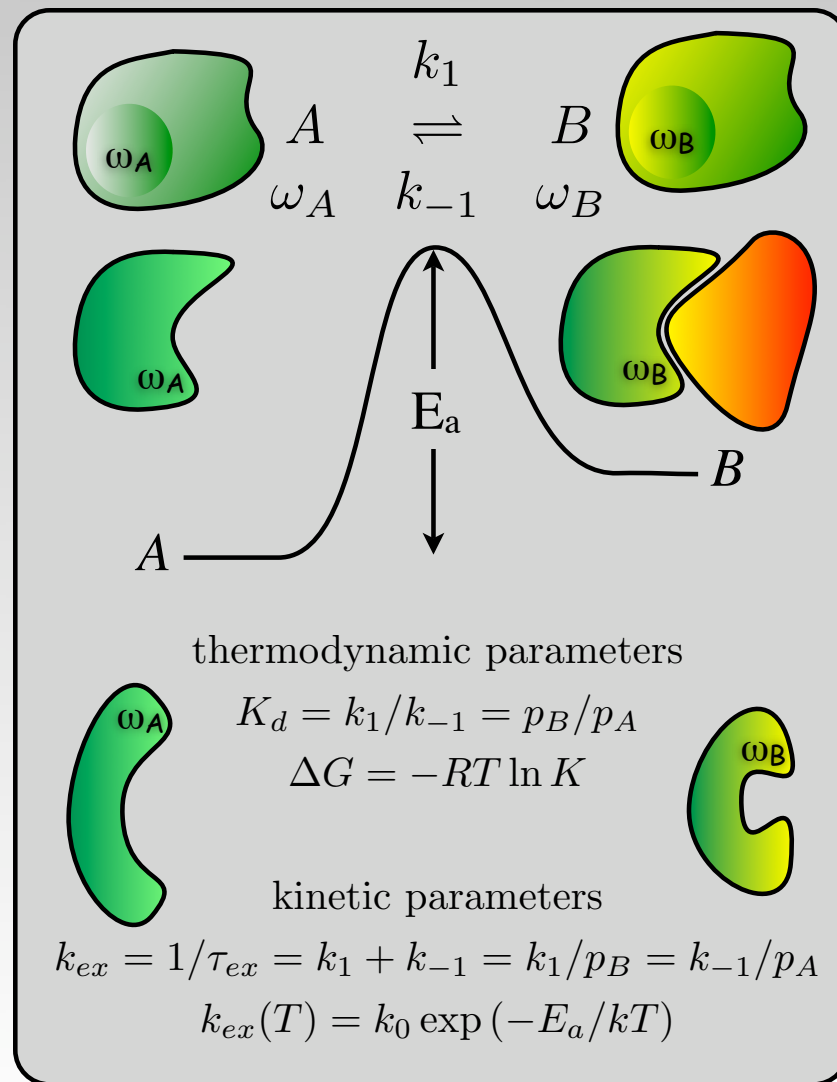
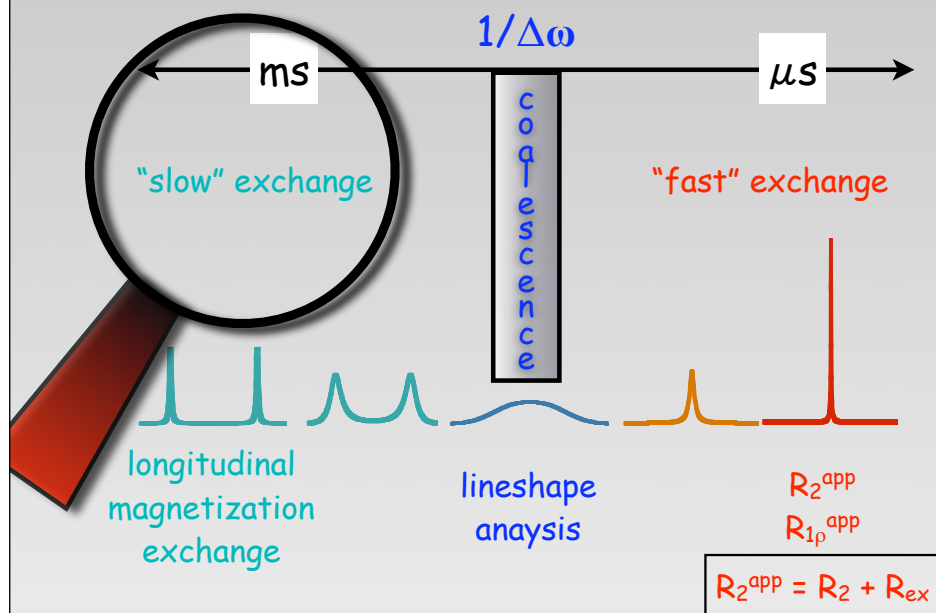
two state non symmetrical exchange

An extreme case : amide protons exchange with water



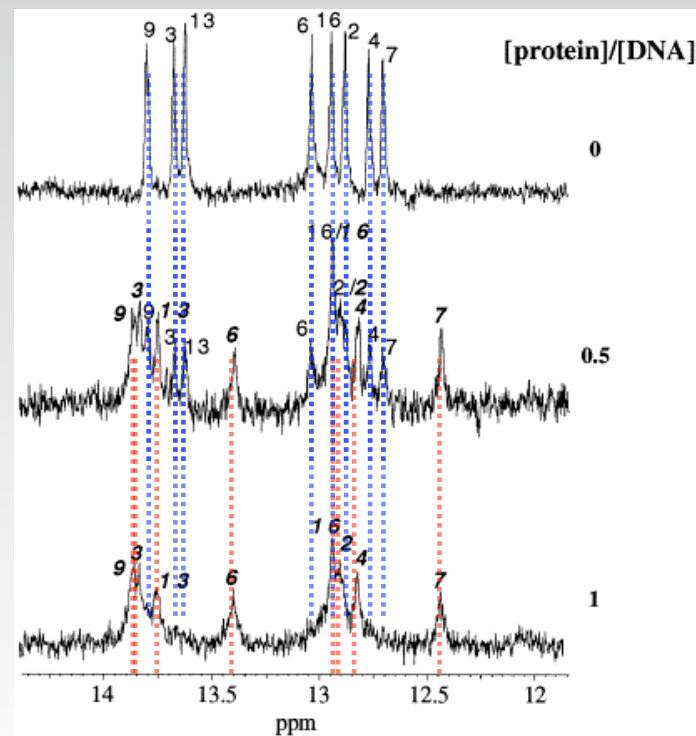
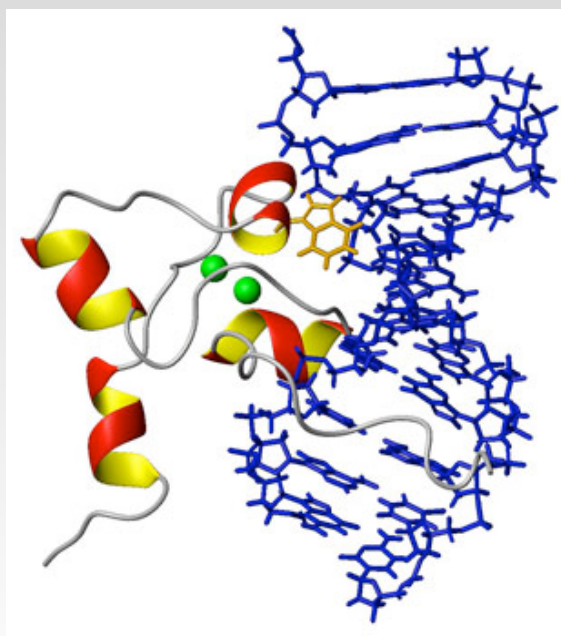
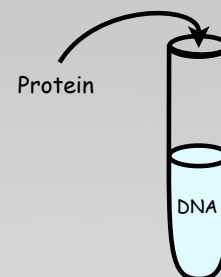
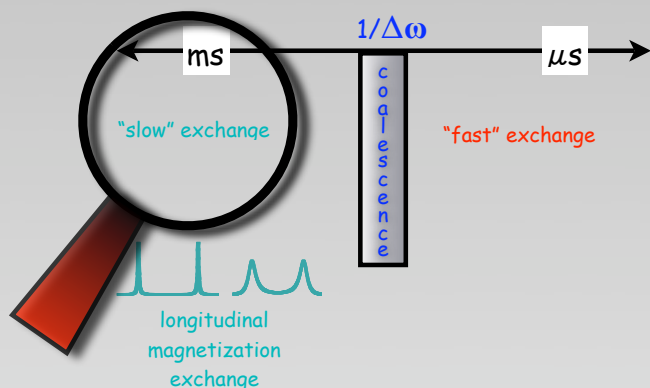
# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

NMR methods to characterize conformational exchange



# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

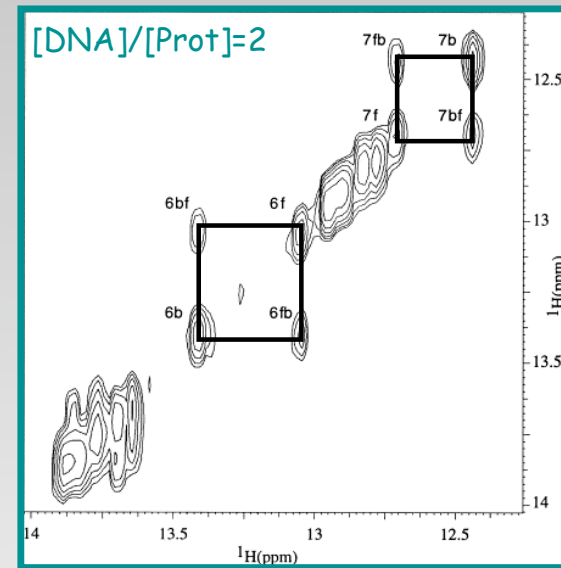
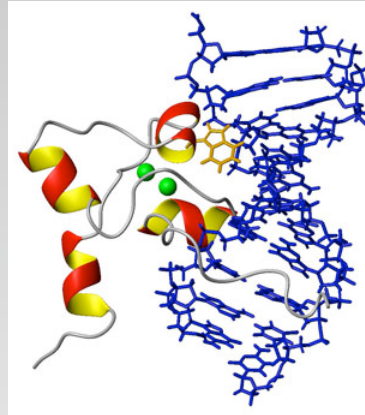
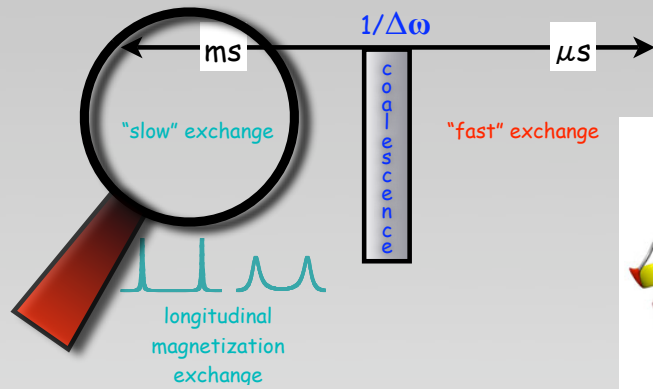
NMR methods to characterize conformational exchange



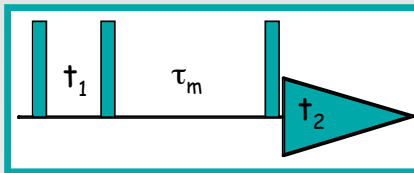


# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

## NMR methods to characterize conformational exchange



$$\begin{bmatrix} \Delta M_{Az}(t) \\ \Delta M_{Bz}(t) \end{bmatrix} = \begin{bmatrix} a_{AA}(t) & a_{AB}(t) \\ a_{BA}(t) & a_{BB}(t) \end{bmatrix} \begin{bmatrix} \Delta M_{Az}(0) \\ \Delta M_{Bz}(0) \end{bmatrix}$$

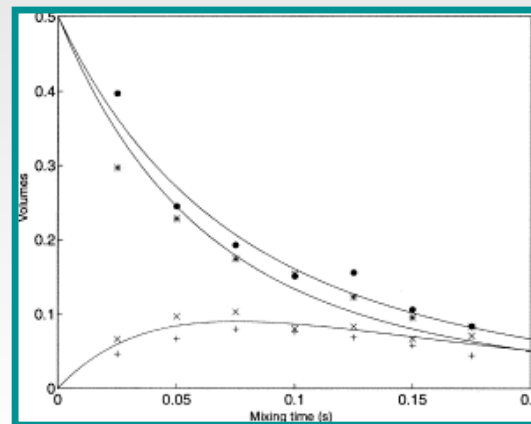


$$V_b(\tau_m) = x_b e^{-\sigma\tau_m} \left[ \cosh(D\tau_m) - \frac{\delta}{D} \sinh(D\tau_m) \right]$$

$$V_f(\tau_m) = x_f e^{-\sigma\tau_m} \left[ \cosh(D\tau_m) + \frac{\delta}{D} \sinh(D\tau_m) \right]$$

$$V_{\text{crosspeak}}(\tau_m) = x_b x_f \frac{k}{D} e^{-\sigma\tau_m} \sinh(D\tau_m)$$

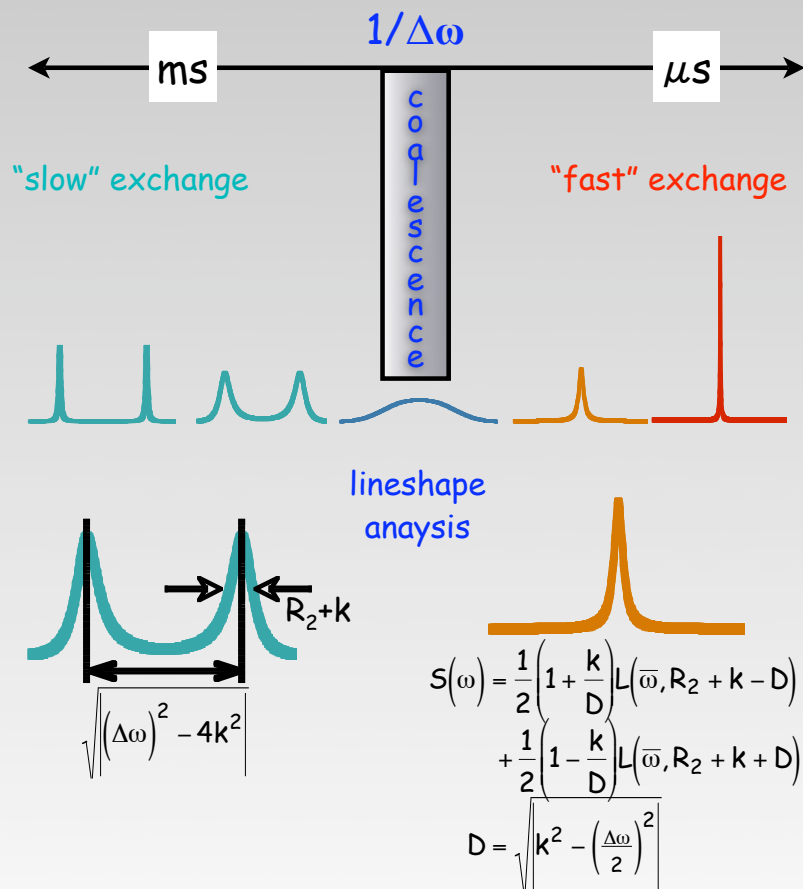
$$\sigma = \frac{1}{2}(R_{1b} + R_{1f}); \delta = \frac{1}{2}(R_{1b} - R_{1f}); D = \sqrt{\delta^2 + x_b x_f k^2}$$



$k_{\text{ex}} = 14 \pm 2 \text{ Hz}$   
 $\tau_{\text{ex}} = 74 \pm 7 \text{ ms}$

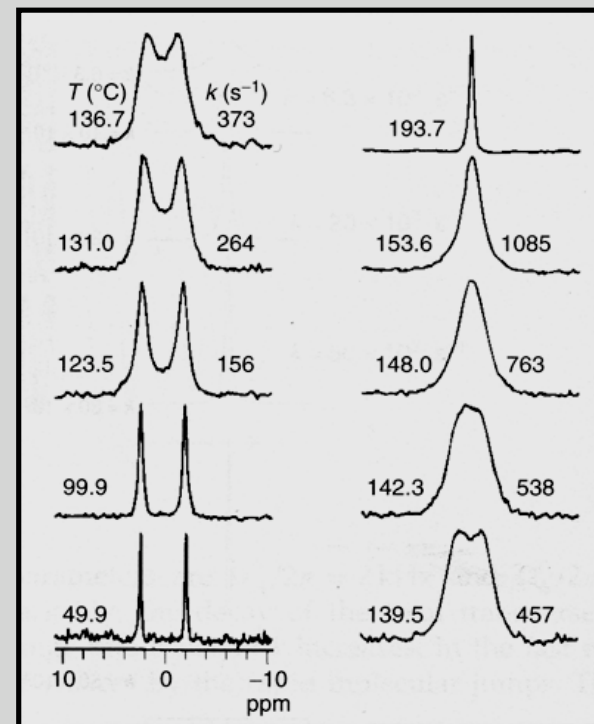
# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

NMR methods to characterize conformational exchange



➔  $k(T) = k_0 \exp\left(-\frac{E_a}{R_B T}\right)$

Perturbation of resonances shape : modification of the exchange regime with temperature



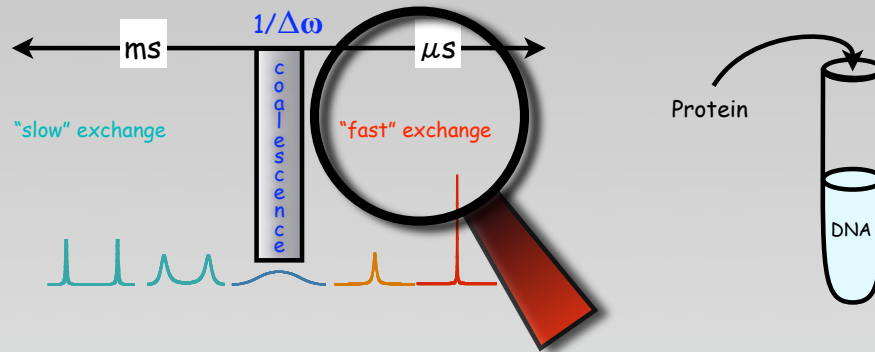
$\{^{13}\text{C}\}$  N,N' diméthylformamide gaz,  $B_0=4,7\text{T}$

$E_a=90,1\text{kJ}\cdot\text{mol}^{-1}$  ;  $k_0=1,16\cdot 10^4\text{Hz}$

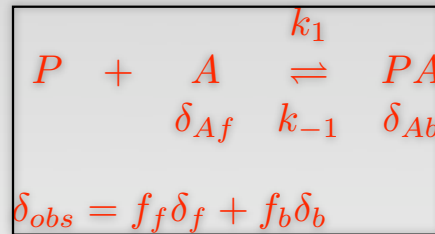
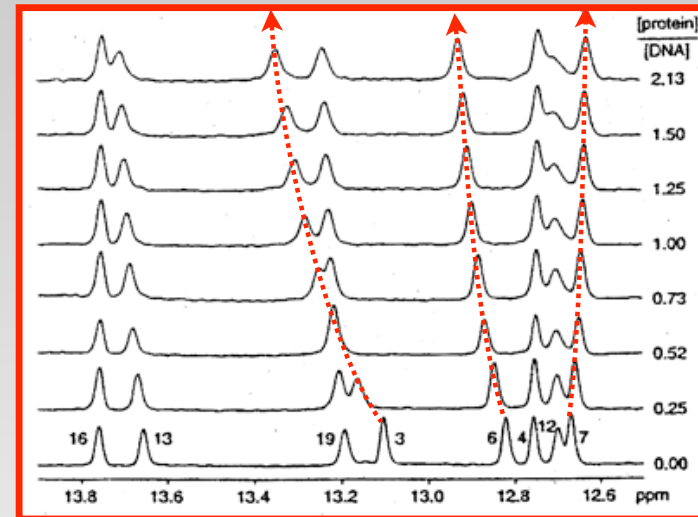
Ross & True, J. Am. Chem. Soc. 106, 2451 (1984)

# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

## NMR methods to characterize conformational exchange



⇒ peaks displacements

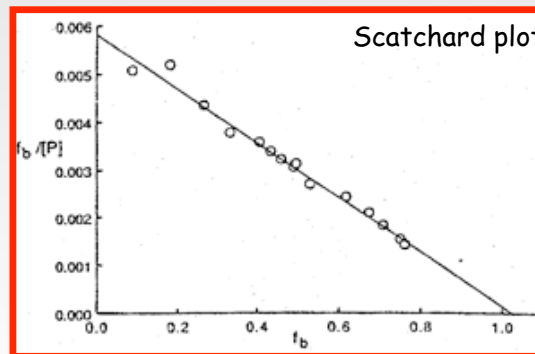


$$K_d = \frac{k_{on}}{k_{off}} = \frac{[P][N]}{[PN]}$$

$$K_d = \frac{(P_0 - f_b N_0)(N_0 - f_b N_0)}{f_b N_0}$$

$$f_b = \frac{\Delta\delta}{\delta_b - \delta_f}$$

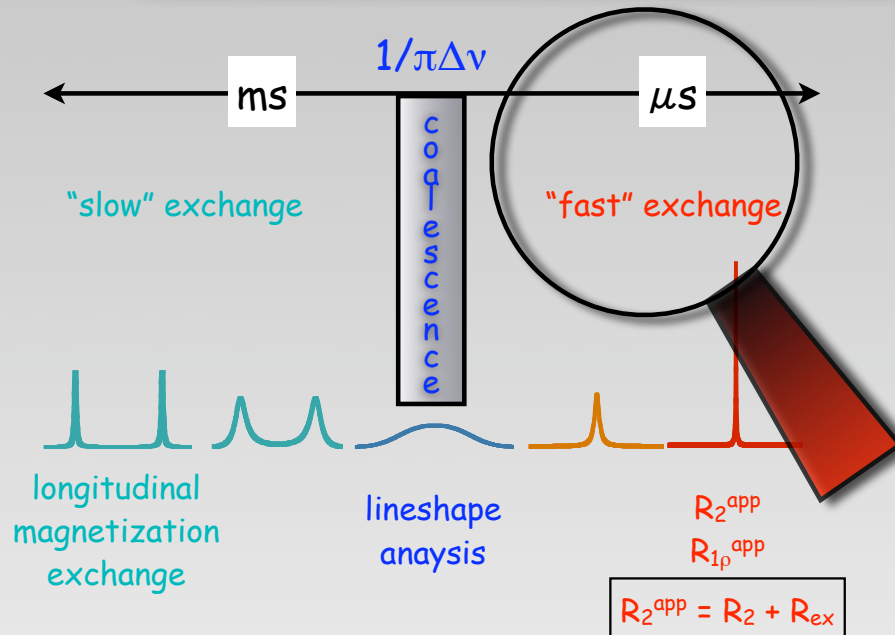
$$\frac{f_b}{[P]} = -\frac{1}{K_d} f_b + \frac{1}{K_d}$$



⇒  $K_d \sim 170 \mu\text{M}$

# Analysis of $\mu\text{s}$ - $\text{ms}$ motions by NMR

for the characterization of exchange processes in proteins



⇒ Case of folding ↔ unfolding processes

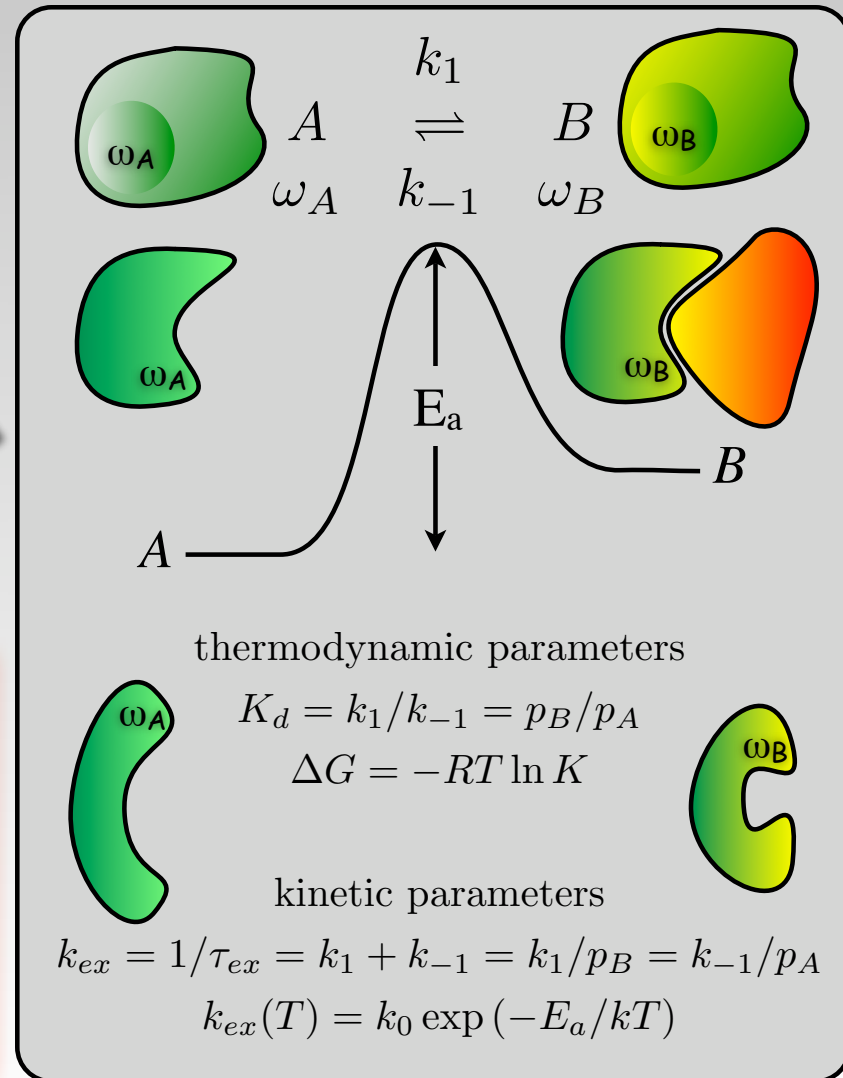
⇒ sequence/stability/motions relationship

⇒ analysis of minor populations

⇒ Motions and function

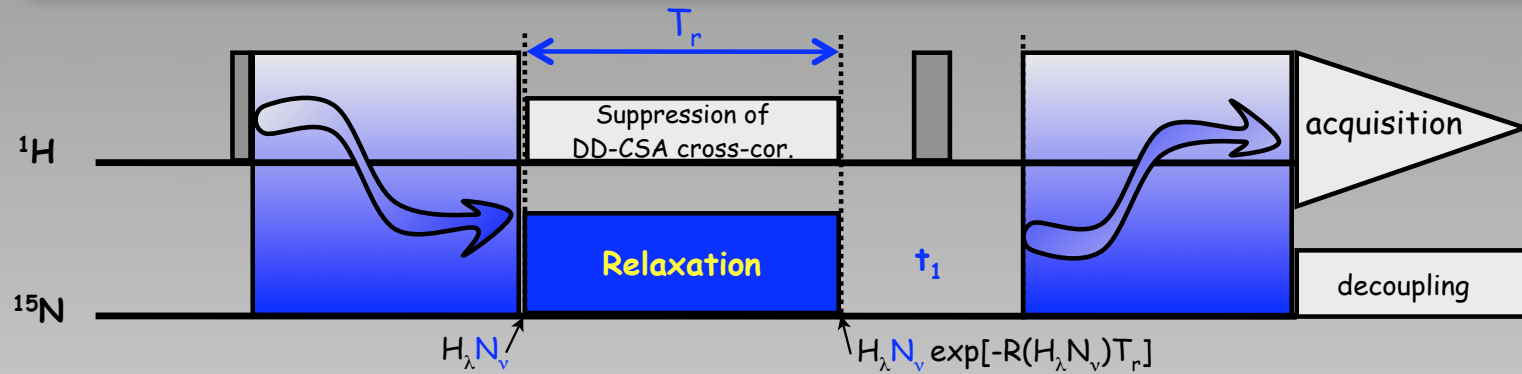
⇒ enzymes cycle

⇒ interactions



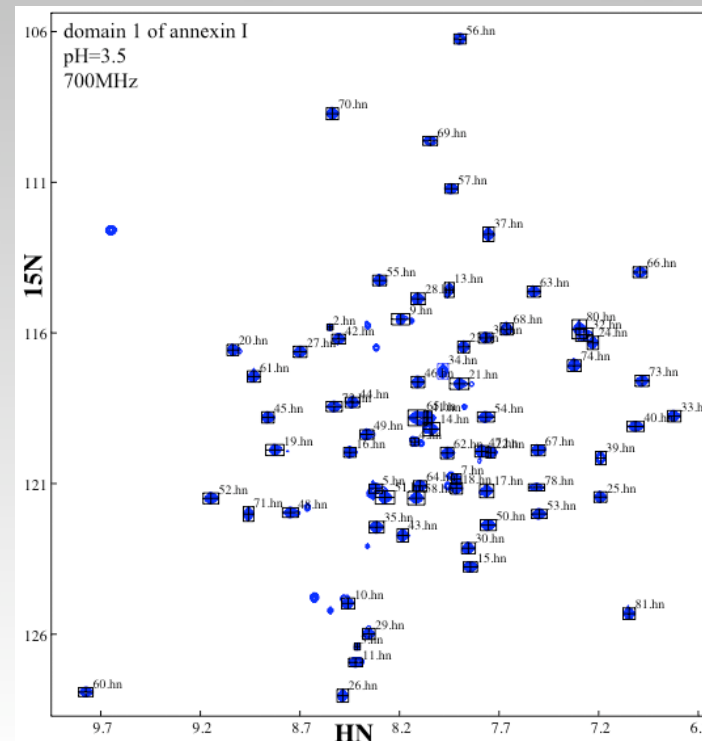
# Characterization of $\mu\text{s}$ -ms motions by NMR

## HOW ?

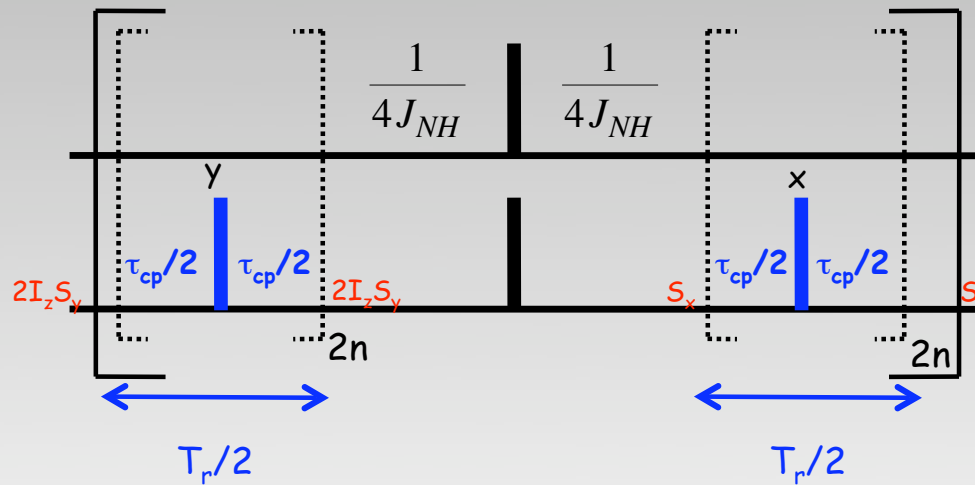
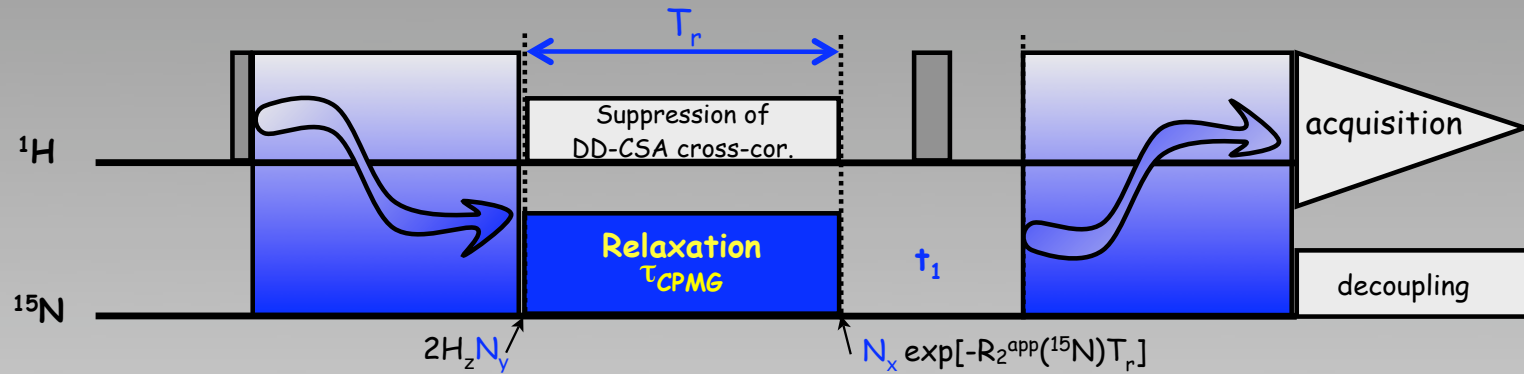


- ⇒ A sample for NMR
- ⇒ concentration  $\sim 0.5\text{mM}$
- ⇒ stable
- ⇒  $^{15}\text{N}$  labeled

- ⇒ An experiment specifically designed for the problem.
- ⇒ Assigned  $^1\text{H}^{\text{N}}$ ,  $^{15}\text{N}$  resonances



# The method : relaxation dispersion NMR experiments



$$R_{2A}^{\text{measured}} = \bar{R} + R_{\text{ex}}(p_A, p_B, k_{\text{ex}}, \Delta\omega, \tau_{\text{CPMG}})$$

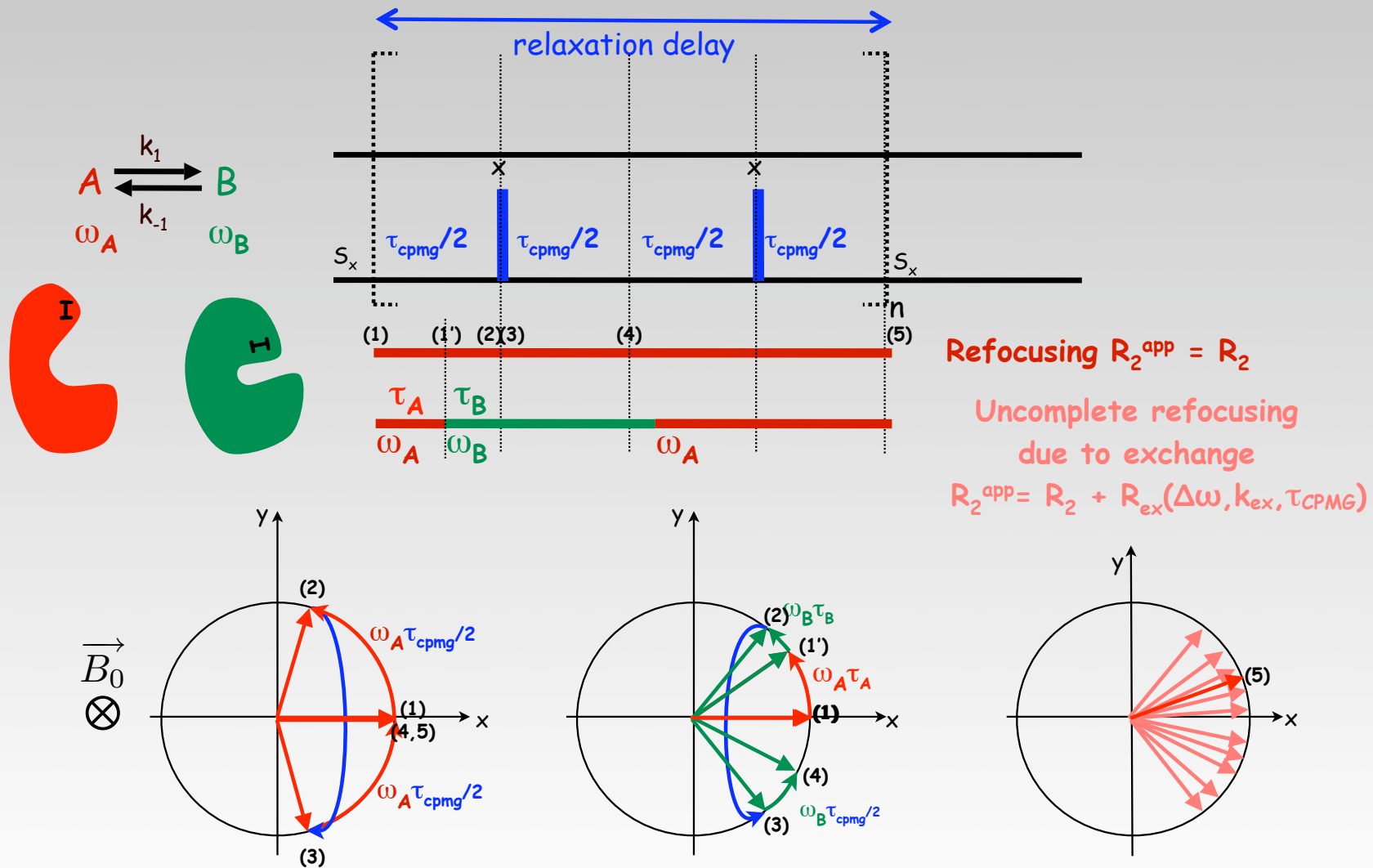
$$\bar{R} = \frac{1}{2}(R_{\text{in}}^0 + R_{\text{anti}}^0)$$

Mulder, F., van Tilborg, P. J. A., Kaptein, R. and Boelens, R. (1999) J. Biomol NMR 13, 275

Deuxième école RMN du CERM, Caen, 2008

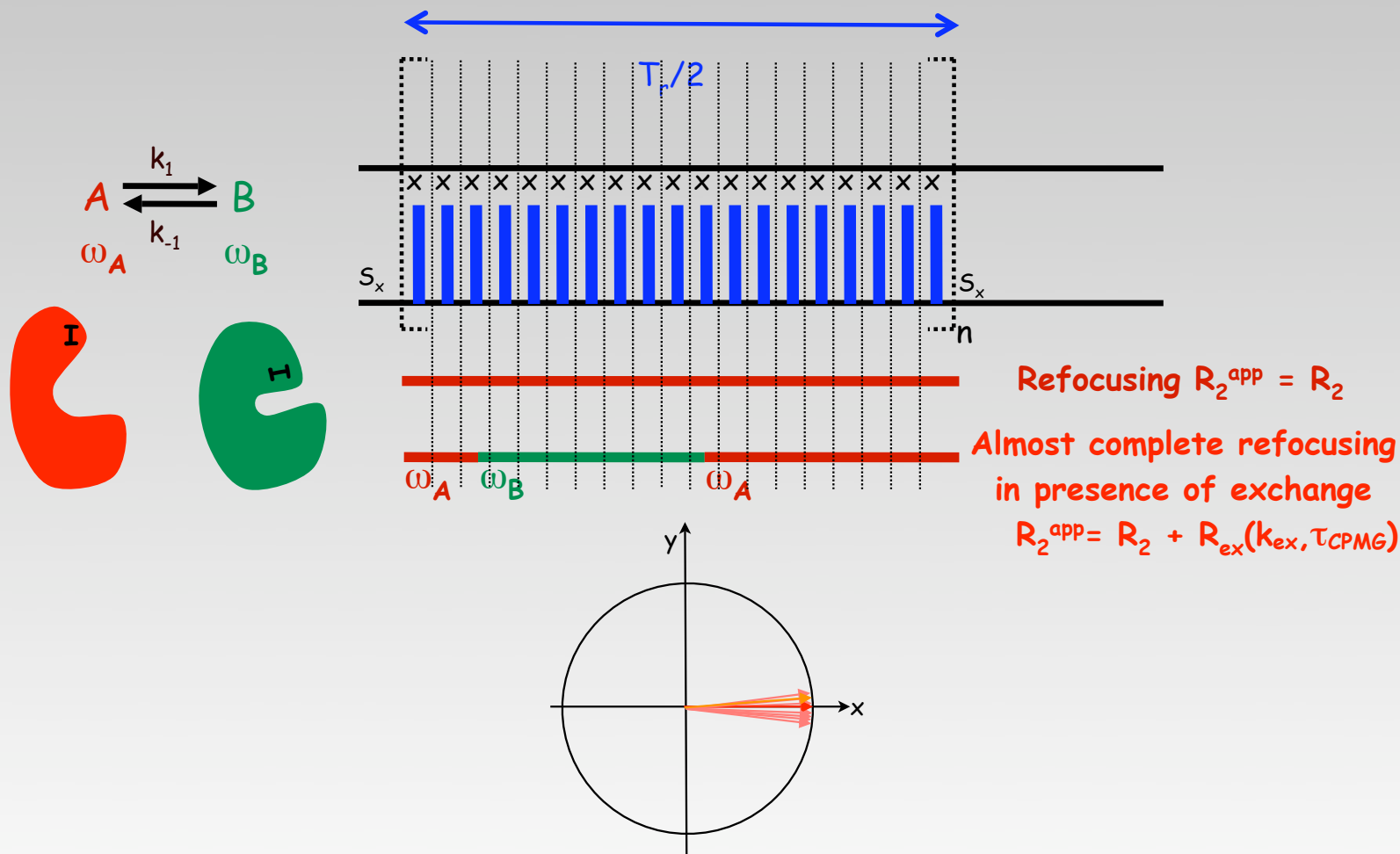
# The method : relaxation dispersion NMR experiments

The principle : variation of  $R_2^{\text{app}}$  as a function of an experimental parameter :  $\tau_{\text{CPMG}}$



# The method : relaxation dispersion NMR experiments

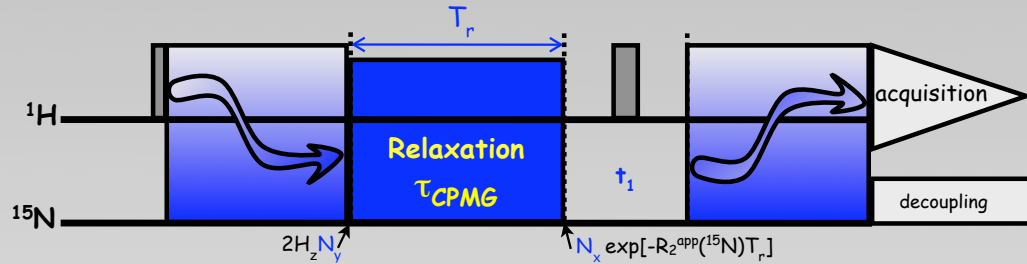
The principle : variation of  $R_2^{\text{app}}$  as a function of an experimental parameter :  $\tau_{\text{CPMG}}$





# The method : relaxation dispersion NMR experiments

The principle : variation of  $R_2^{app}$  as a function of an experimental parameter :  $\tau_{CPMG}$



$$R_2^{measured} = R_2^0 + R_{ex}(p_A, p_B, k_{ex}, \Delta\omega, \tau_{CPMG})$$

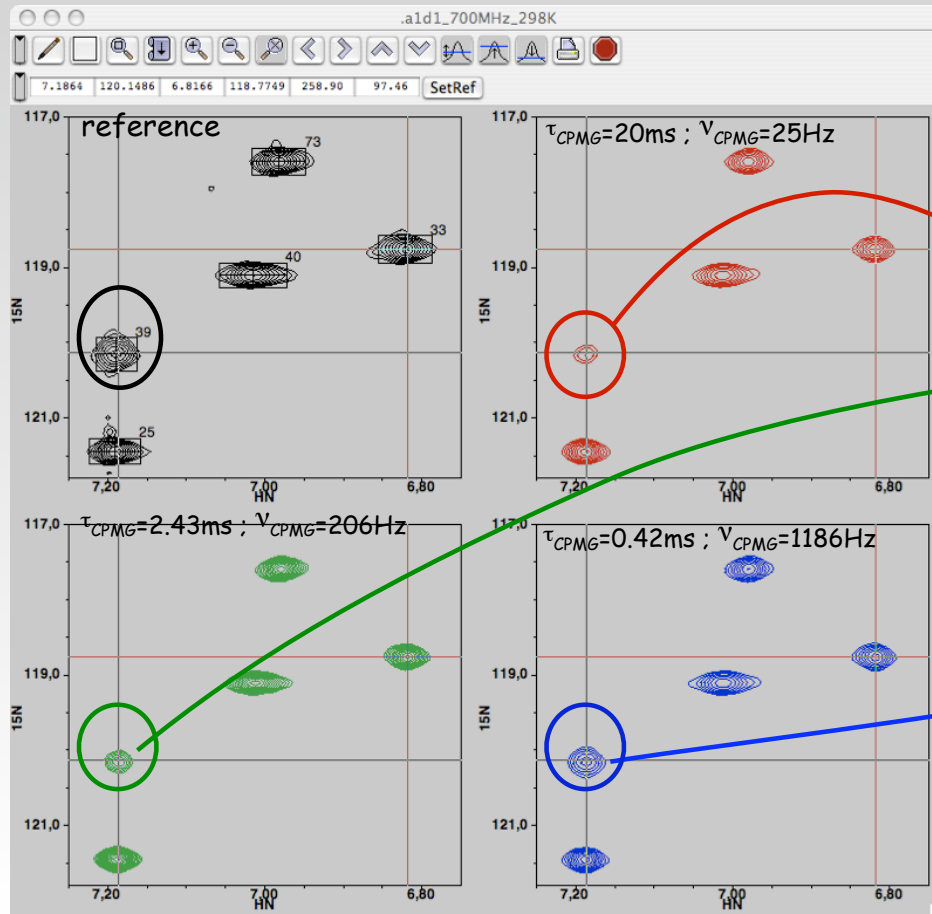
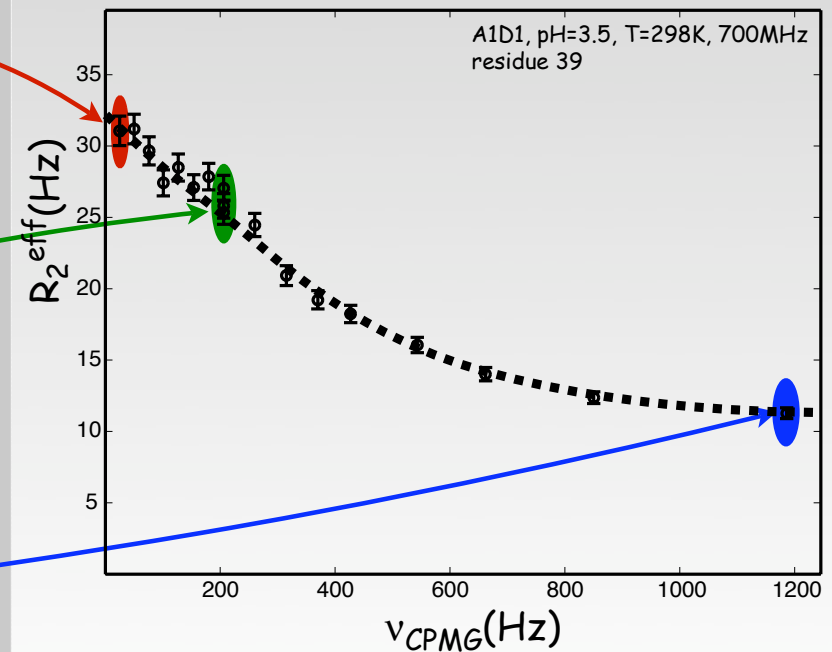
$$\frac{I(T_r)}{I_0} = \exp[-(R_2^0 + R_{ex})T_r]$$

$$R_2^{measured}(\nu_{CPMG}) = -\frac{1}{T_r} \ln \left[ \frac{I(T_r, \nu_{CPMG})}{I_0} \right]$$

$$T_r = 2n\tau_{CPMG} = \frac{n}{\nu_{CPMG}}$$

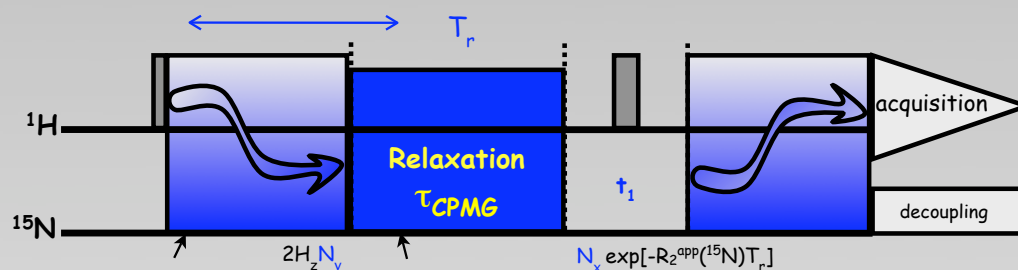
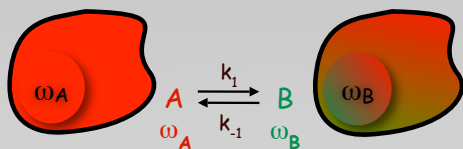
Relaxation dispersion curve

$$R_2^{eff} = f(\nu_{CPMG})$$



# The method : relaxation dispersion NMR experiments

First analysis : qualitative analysis of relaxation dispersion curves



$$R_2^{measured} = R_2^0 + R_{ex}(p_A, p_B, k_{ex}, \Delta\omega, \tau_{CPMG})$$

$$\frac{I(T_r)}{I_0} = \exp[-(R_2^0 + R_{ex})T_r]$$

$$R_2^{measured}(\nu_{CPMG}) = -\frac{1}{T_r} \ln \left[ \frac{I(T_r, \nu_{CPMG})}{I_0} \right]$$

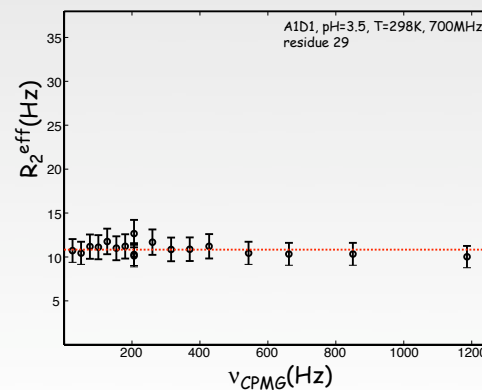
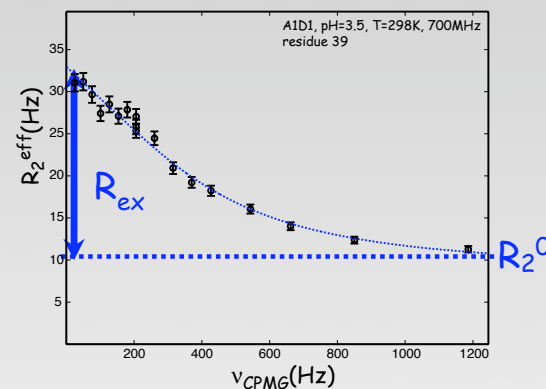
$$T_r = 2n\tau_{CPMG} = \frac{n}{\nu_{CPMG}}$$

⇒ First estimation of  $R_{ex}$

$$R_{ex}(\text{estimated}) = R_2^{eff}(\nu_{min}) - R_2^{eff}(\nu_{max})$$

⇒ Fitting of the dispersion curve

- ⇒  $R_{ex}, R_2^0$
- ⇒ structural parameter  $\Delta\omega$
- ⇒ thermodynamic parameter  $p_A, p_B$
- ⇒ kinetic parameter :  $k_{ex} = 1/\tau_{ex}$
- ⇒  $k_{ex}(T) = k_0 \exp(-E_a/R_B T)$



# The method : relaxation dispersion NMR experiments

## First analysis : qualitative analysis of relaxation dispersion curves

⇒ First estimation of  $R_{ex}$

$$R_{ex}(\text{estimated}) = R_2^{\text{eff}}(v_{\min}) - R_2^{\text{eff}}(v_{\max})$$

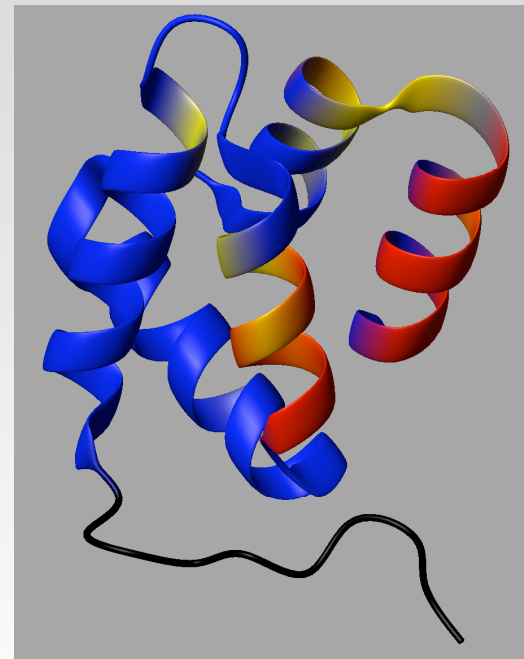
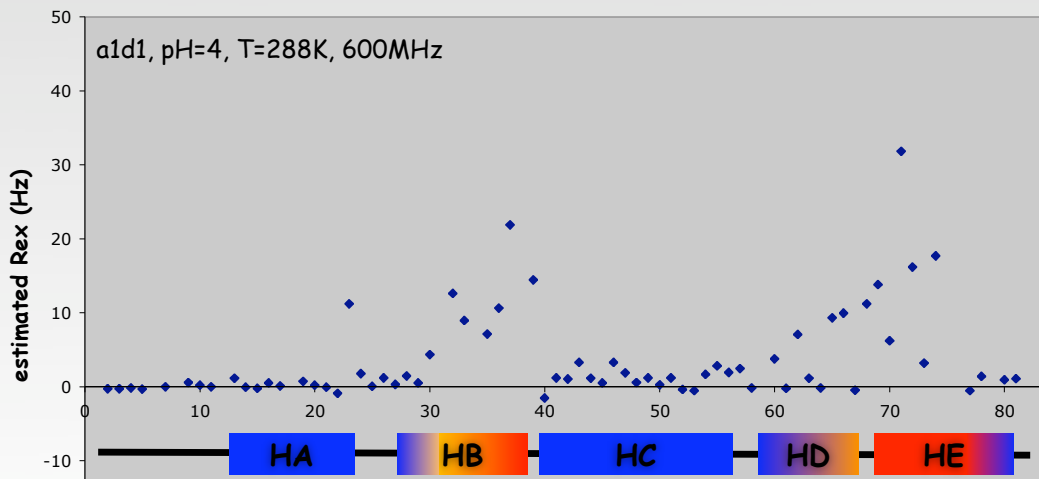
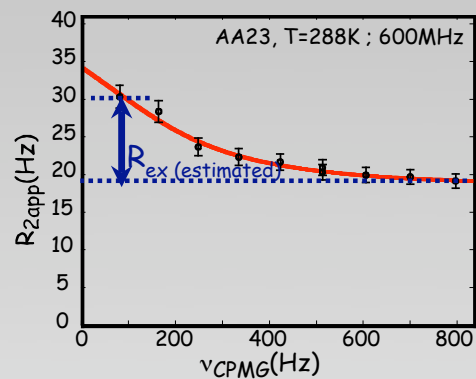
⇒ Fitting of the dispersion curve

⇒  $R_{ex}, R_2^0$

⇒  $\Delta\omega, p_A$

⇒  $k_{ex} = 1/\tau_{ex}$

⇒  $k_{ex}(T) = k_0 \exp(-E_a/R_B T)$

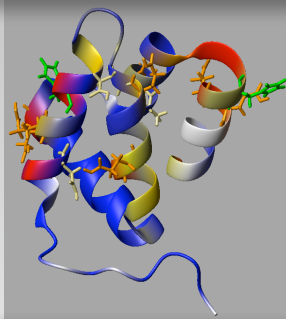


# The method : relaxation dispersion NMR experiments

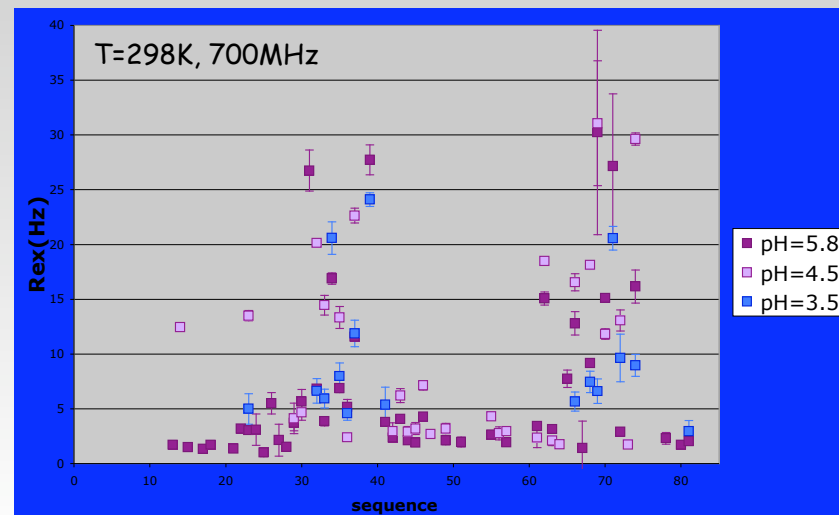
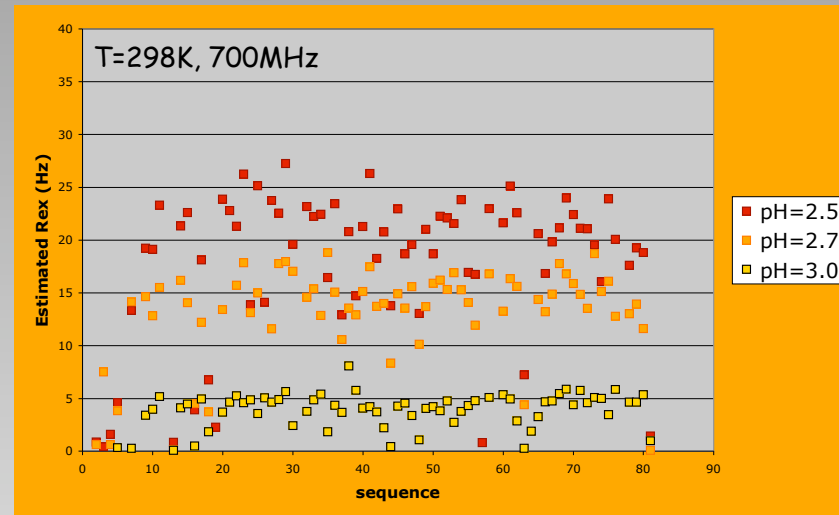
First analysis : qualitative analysis of relaxation dispersion curves

An example : variation of exchange contribution as a function of pH

pH < 3  
Global conformational exchange  
☞ Global Unfolding ?



pH > 3.5  
Localized conformational exchange  
☞ Local Unfolding ?



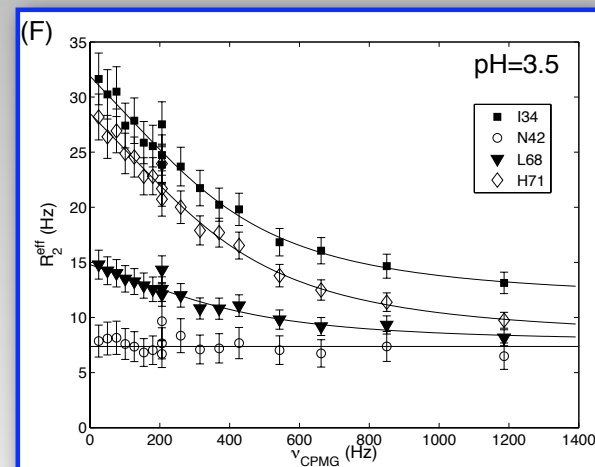
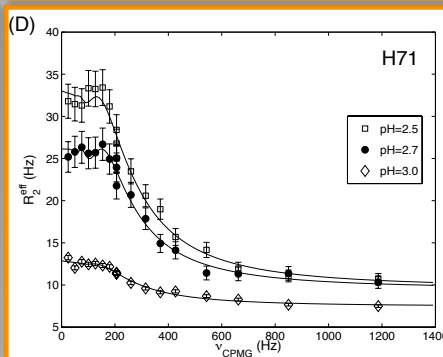
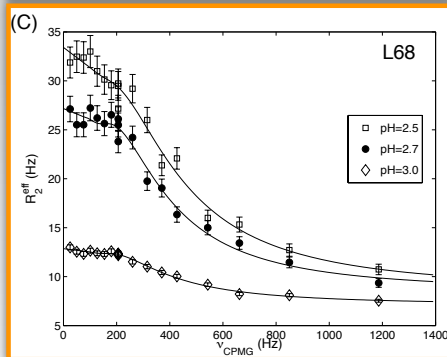
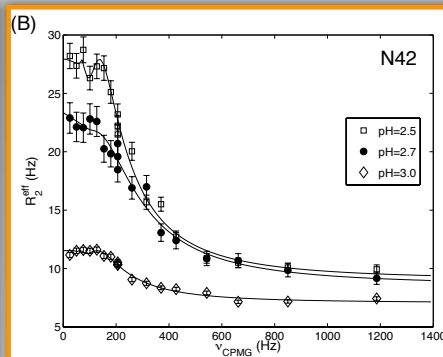
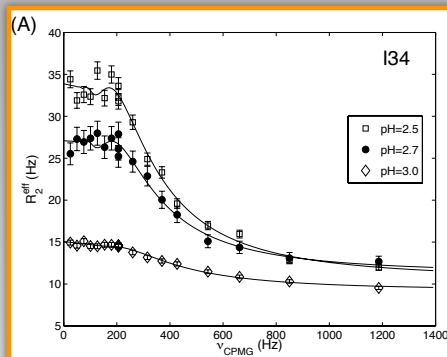
# The method : relaxation dispersion NMR experiments

First analysis : qualitative analysis of relaxation dispersion curves

An example : variation of exchange contribution as a function of pH

pH < 3  
Global conformational exchange

pH > 3.5  
Localized conformational exchange



👉 Physical parameters of these exchange processes ?

# The method : relaxation dispersion NMR experiments

Quantitative evaluation of physical parameters of exchange from the relaxation dispersion data

The analysis depends of the timescale of the process

**Fast exchange** :  $\Delta\omega \ll k_{ex}$

- ⇒ One peak  $\omega = p_A\omega_A + p_B\omega_B$
- ⇒  $R_{ex} \propto B_0^2$
- ⇒ Meiboom-Gill equation
- ⇒  $k_{ex}, \Phi = p_A p_B \Delta\omega^2, R_{inf}$

**Intermediate exchange** :  $\Delta\omega \sim k_{ex}$

- ⇒ One or two broadened peaks
- ⇒  $R_{ex} \propto B_0$
- ⇒ Carver-Richards equation
- ⇒  $k_{ex}, p_B, \Delta\omega, R_{inf}$

**Slow exchange** :  $\Delta\omega \gg k_{ex}$

- ⇒ Two peaks
- ⇒  $R_{ex}$  independant of  $B_0$
- ⇒ Tollinger equation
- ⇒  $k_a, \Delta\omega, R_{inf}$
- ⇒  $R_{ex}^{(B)} \rightarrow k_{BA} = p_A k_{ex}$  ;  $R_{ex}^{(A)} \rightarrow k_{AB} = p_B k_{ex}$   
 $R_{ex}^{(B)} \gg R_{ex}^{(A)}$

Detection of minor populations from the broadening of the major state resonance : use of Carver-Richards equation

$$p_A=0.2 ; p_B=0.8$$

$$R_2^{app} = R_2^{inf} + p_A p_B \Delta\omega^2 \tau_{ex} \left[ 1 - 4\tau_{ex} \nu_{CP} \tanh \frac{1}{4\tau_{ex} \nu_{CP}} \right]$$

$\frac{1}{k_{ex}/\Delta\nu}$

$$R_2^{app} = R_2^{inf} + 0.5k_{ex} - \nu_{CP} a \cosh(D_+ \cosh \eta_+ - D_- \cos \eta_-)$$

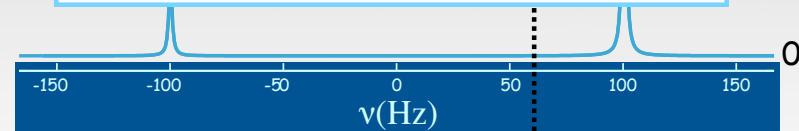
$$D_+ = 0.5 \left[ 1 + \frac{\phi + 2\Delta\omega^2}{\sqrt{\phi^2 + \zeta^2}} \right] ; D_- = 0.5 \left[ -1 + \frac{\phi + 2\Delta\omega^2}{\sqrt{\phi^2 + \zeta^2}} \right]$$

$$\eta_+ = \frac{\tau_{CP}}{\sqrt{(2)}} \sqrt{\phi + \sqrt{\phi^2 + \zeta^2}} ; \eta_- = \frac{\tau_{CP}}{\sqrt{(2)}} \sqrt{-\phi + \sqrt{\phi^2 + \zeta^2}}$$

$$\phi = [(p_B - p_A)k_{ex}]^2 - \Delta\omega^2 + 4p_A p_B k_{ex}^2$$

$$\zeta = 2\Delta\omega(p_B - p_A)k_{ex}$$

$$R_2^{app} = R_2^{inf} + k_a \left( 1 - \frac{\sin \Delta\omega \cdot \tau_{CP}}{\Delta\omega \cdot \tau_{CP}} \right)$$



$$p_A \nu_A + p_B \nu_B$$

# The method : relaxation dispersion NMR experiments

## Quantitative evaluation of physical parameters of exchange from the relaxation dispersion data

The analysis depends of the timescale of the process

### Fast exchange : $\Delta\omega \ll k_{ex}$

- ⇒ One peak  $\omega = p_A\omega_A + p_B\omega_B$
- ⇒  $R_{ex} \propto B_0^2$
- ⇒ Meiboom-Gill equation
- ⇒  $k_{ex}, \Phi = p_A p_B \Delta\omega^2, R_{inf}$

$$R_2^{app} = R_2^{inf} + p_A p_B \Delta\omega^2 \tau_{ex} \left[ 1 - 4\tau_{ex} \nu_{CP} \tanh \frac{1}{4\tau_{ex} \nu_{CP}} \right]$$

### Intermediate exchange : $\Delta\omega \sim k_{ex}$

- ⇒ One or two broadened peaks
- ⇒  $R_{ex} \propto B_0$
- ⇒ Carver-Richards equation
- ⇒  $k_{ex}, p_B, \Delta\omega, R_{inf}$

$$R_2^{app} = R_2^{inf} + 0.5k_{ex} - \nu_{CP} a \cosh(D_+ \cosh \eta_+ - D_- \cos \eta_-)$$

$$D_+ = 0.5 \left[ 1 + \frac{\phi + 2\Delta\omega^2}{\sqrt{\phi^2 + \zeta^2}} \right]; D_- = 0.5 \left[ -1 + \frac{\phi + 2\Delta\omega^2}{\sqrt{\phi^2 + \zeta^2}} \right]$$

$$\eta_+ = \frac{\tau_{CP}}{\sqrt{2}} \sqrt{\phi + \sqrt{\phi^2 + \zeta^2}}; \eta_- = \frac{\tau_{CP}}{\sqrt{2}} \sqrt{-\phi + \sqrt{\phi^2 + \zeta^2}}$$

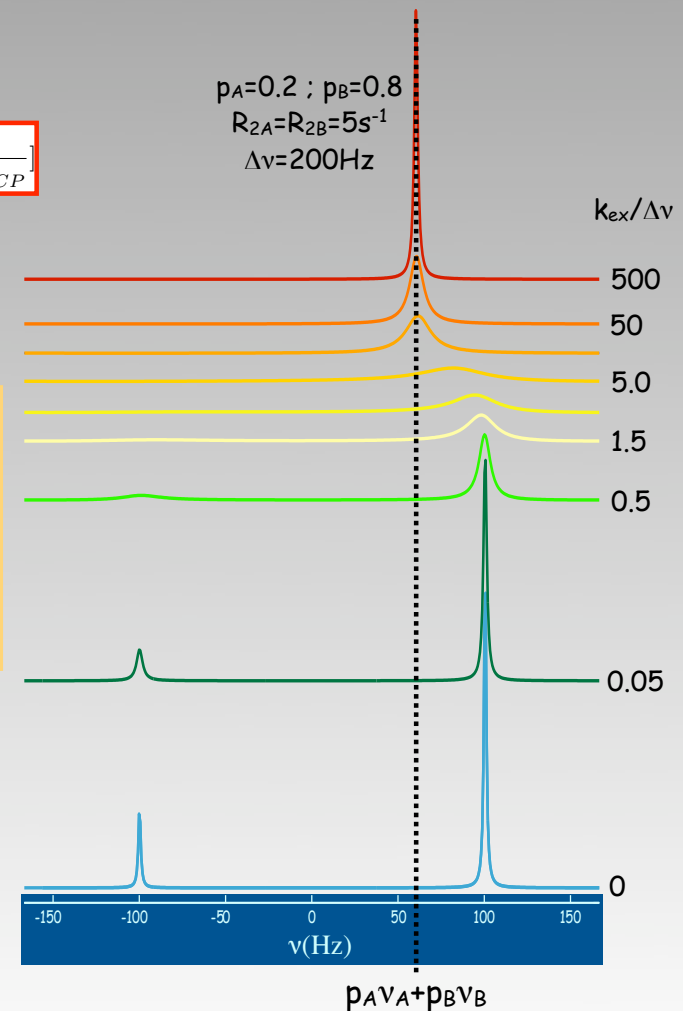
$$\phi = [(p_B - p_A)k_{ex}]^2 - \Delta\omega^2 + 4p_A p_B k_{ex}^2$$

$$\zeta = 2\Delta\omega(p_B - p_A)k_{ex}$$

### Slow exchange : $\Delta\omega \gg k_{ex}$

- ⇒ Two peaks
- ⇒  $R_{ex}$  independant of  $B_0$
- ⇒ Töllinger equation
- ⇒  $k_a, \Delta\omega, R_{inf}$
- ⇒  $R_{ex}^{(B)} \rightarrow k_{BA} = p_A k_{ex}; R_{ex}^{(A)} \rightarrow k_{AB} = p_B k_{ex}$
- ⇒  $R_{ex}^{(B)} \gg R_{ex}^{(A)}$

$$R_2^{app} = R_2^{inf} + k_a \left( 1 - \frac{\sin \Delta\omega \cdot \tau_{CP}}{\Delta\omega \cdot \tau_{CP}} \right)$$

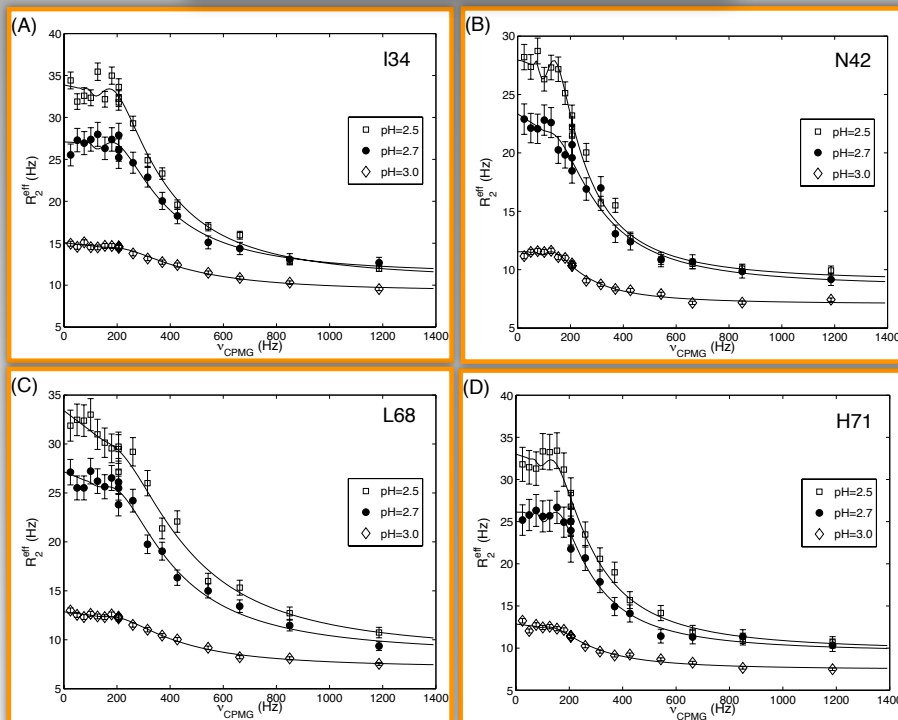


- ⇒ Detection of minor populations from the broadening of the major state resonance : use of Carver-Richards equation

# The method : relaxation dispersion NMR experiments

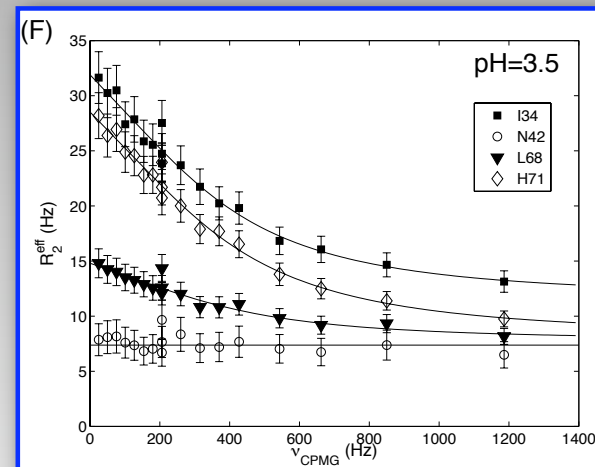
Quantitative evaluation of physical parameters of exchange from the relaxation dispersion data

pH < 3  
Global conformational exchange



Intermediate/Slow exchange regime  
 $p_B \ll p_A$

pH > 3.5  
Localized conformational exchange

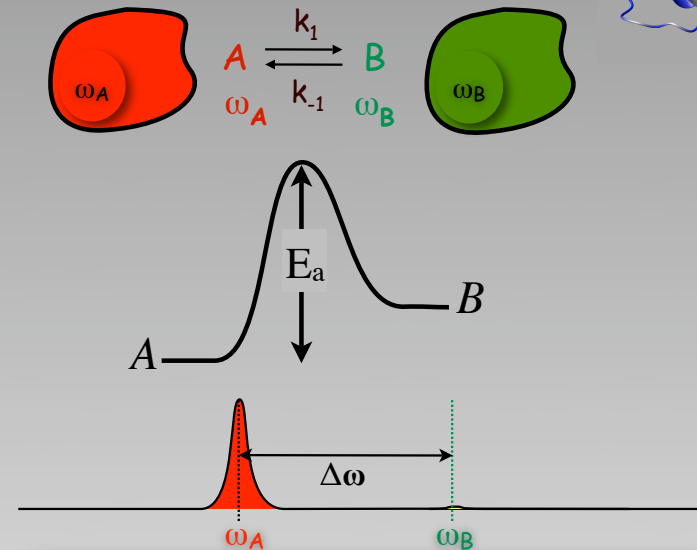
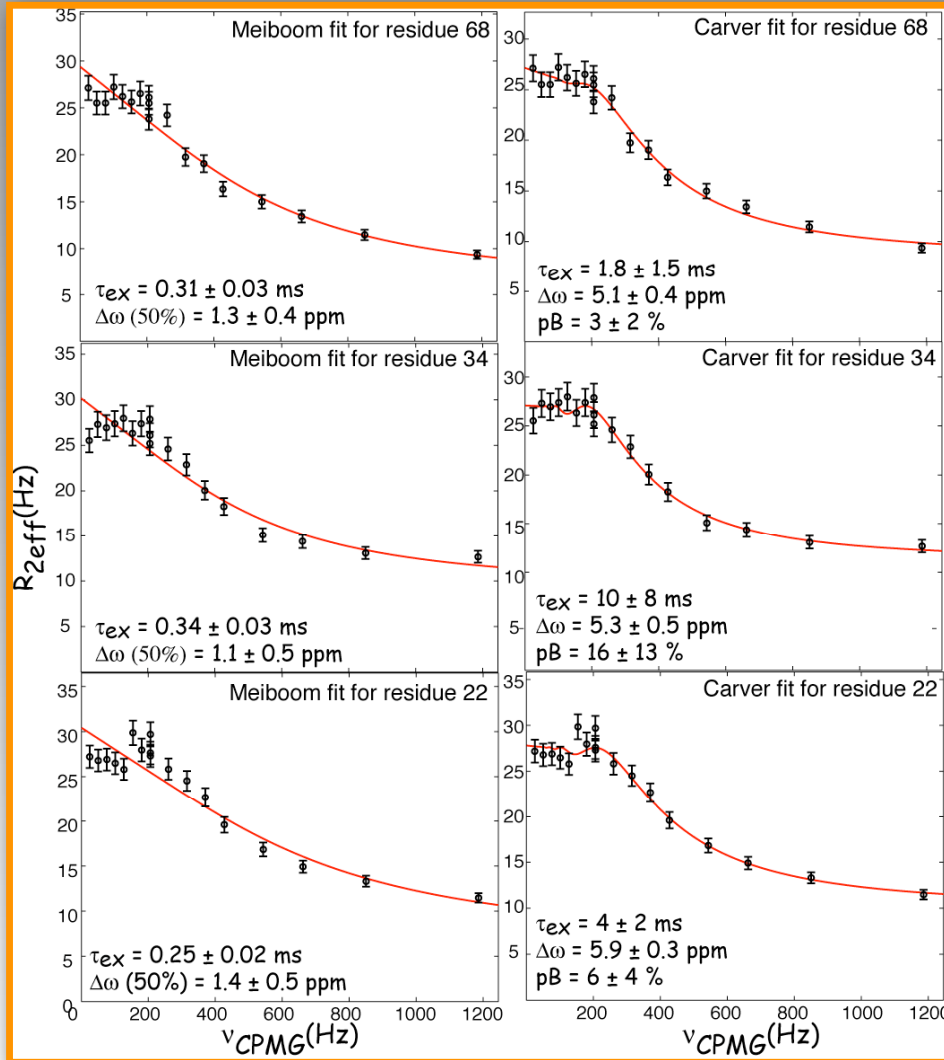
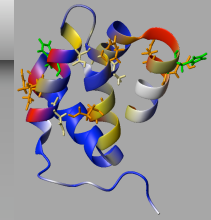


Fast exchange regime  
 $p_B \approx p_A$



# The method : relaxation dispersion NMR experiments

## relaxation dispersion experiments for the analysis of "invisible" states



pH < 3

☞ Individual fits

☞ Carver statistically favored

☞  $\tau_{ex}$  : few ms

☞  $p_B \ll p_A$

☞ Global exchange

☞ global  $\tau_{ex}$ ,  $p_B$

☞ individual  $\Delta\omega$ ,  $R_{inf}$

# The method : relaxation dispersion NMR experiments

relaxation dispersion experiments for the analysis of "invisible" states

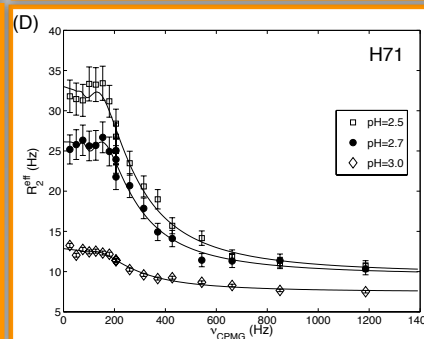
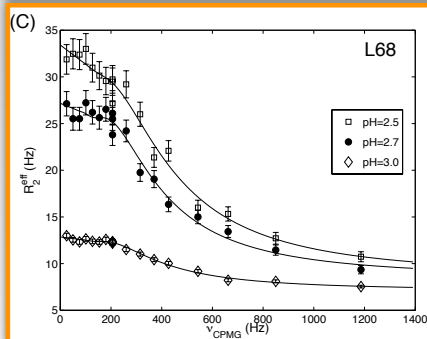
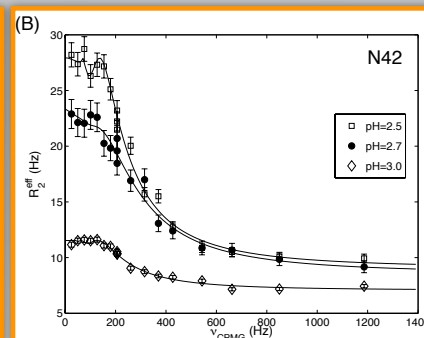
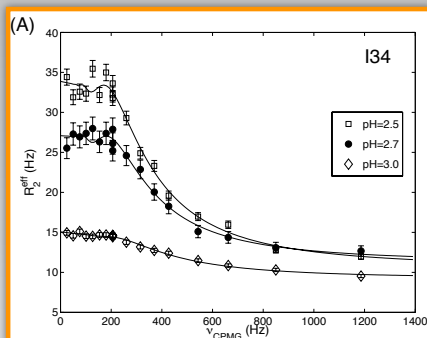
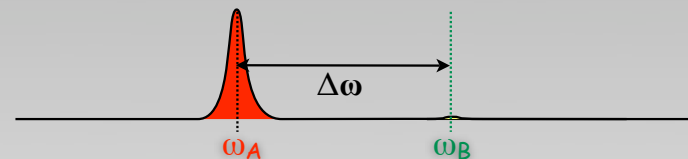
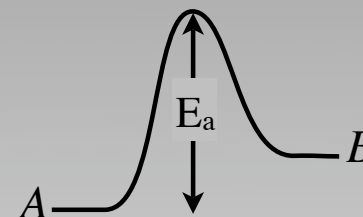
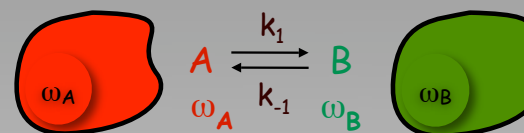
pH < 3

☞ Global exchange

☞  $\tau_{ex} = 4.0 \pm 0.2$  ms (pH=2.5)

☞  $p_B = 1\%$  (pH=3)  $\rightarrow 10\%$  (pH=2.5)

☞ Individual  $\Delta\omega(^{15}\text{N}) = 0 \leftrightarrow 8$  ppm



☞ Broadening of the major component gives information about a minor, invisible state

☞ Nature of the unfolded state ?

# The method : relaxation dispersion NMR experiments

relaxation dispersion experiments for the analysis of "invisible" states

pH < 3

☞ Global unfolding

☞  $\tau_{ex} = 2-5$  ms

☞  $p_B = 1\%$  (pH=3)  $\rightarrow$  10% (pH=2.5)

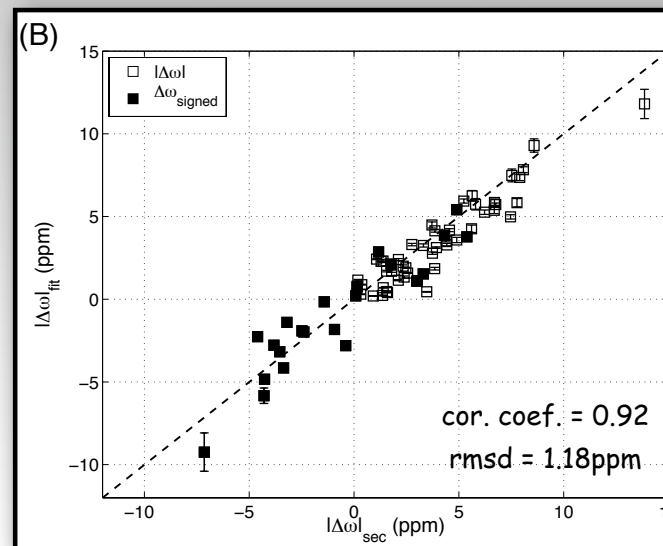
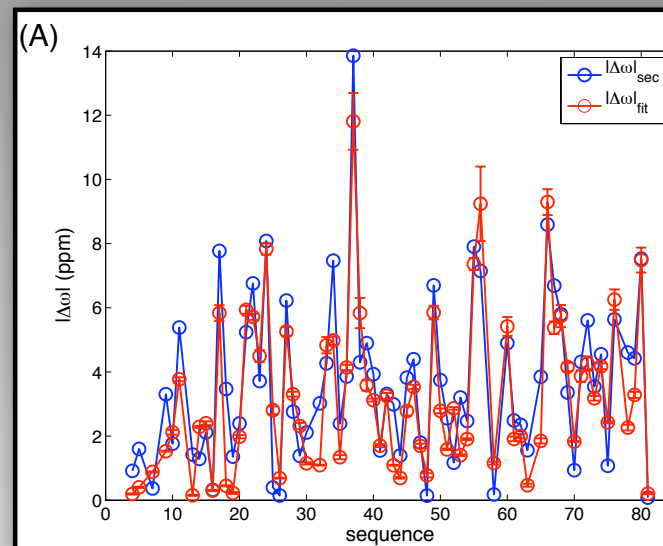
☞  $\Delta\omega(^{15}\text{N}) = 0 \leftrightarrow 8$  ppm

☞ Nature of the unfolded state ?

$\Delta\omega(\text{fitted})$  vs Secondary Shifts

☞ "Fully" unfolded state

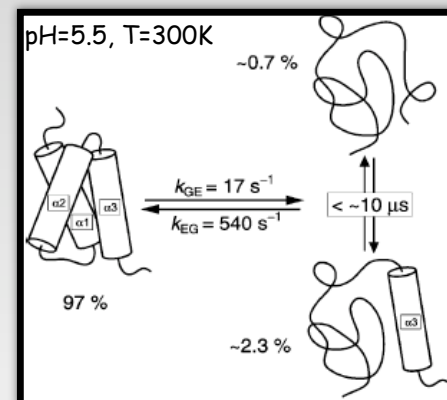
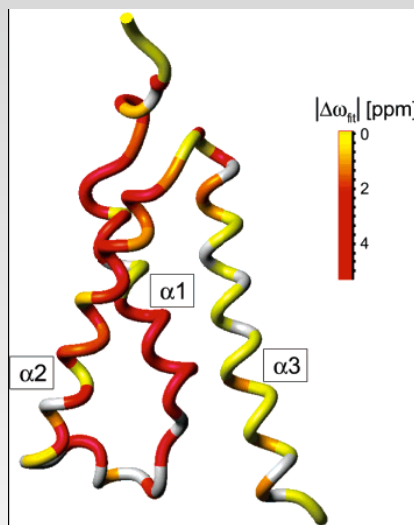
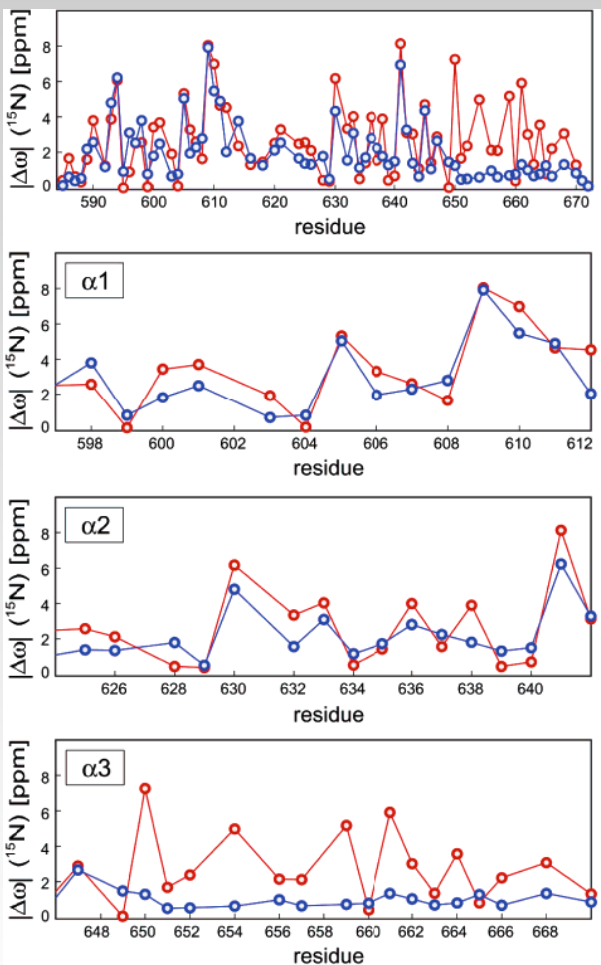
☞ Two-state unfolding process



# The method : relaxation dispersion NMR experiments

relaxation dispersion experiments for the analysis of "invisible" states

## An Isolated Helix Persists in a Sparsely Populated Form of KIX under Native Conditions



# The method : relaxation dispersion NMR experiments

Quantitative evaluation of physical parameters of exchange from the relaxation dispersion data

$3.5 < \text{pH} < 7$

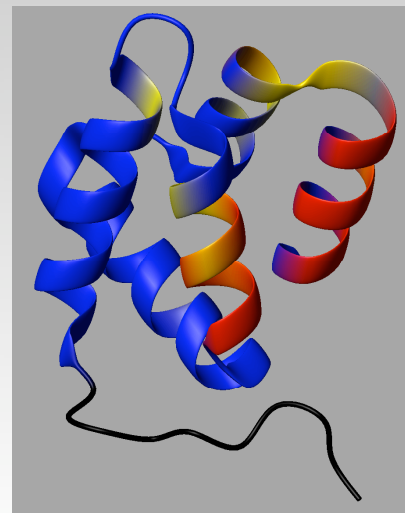
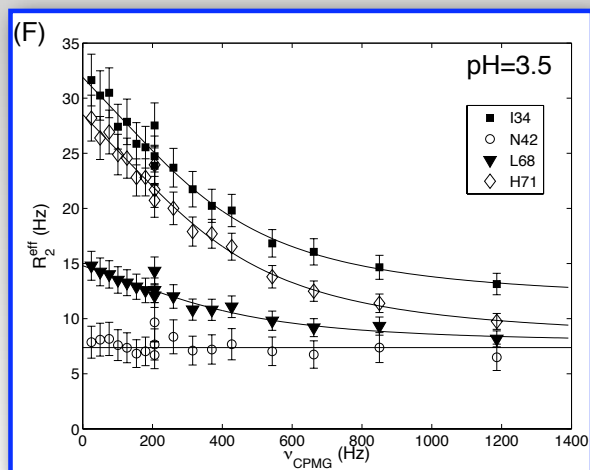
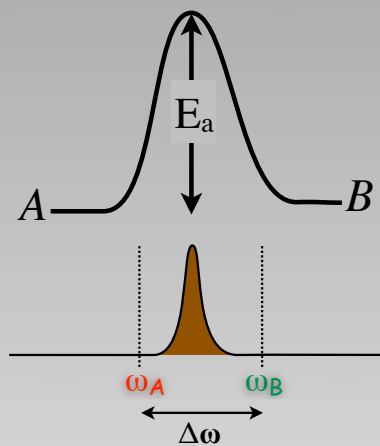
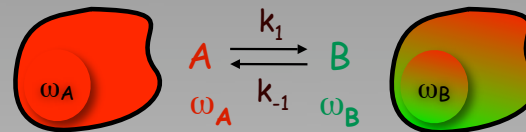
☞ Localized exchange

☞ Meiboom statistically favored  
(fast exchange)

☞  $\tau_{\text{ex}} \sim 0.5 \text{ ms}$  (pH=4.5)

☞  $p_B \approx p_A$

☞  $\Delta\omega(^{15}\text{N}) \sim 0 \leftrightarrow 3 \text{ ppm}$



# $\mu$ S-ms motions by NMR

## conformational exchange processes and proteins function

### ⇒ Folding/unfolding processes

⇒ sequence/stability/flexibility relationship

⇒ low populated states

Transition state ensemble, minor folded/unfolded states

### ⇒ Motions and function

⇒ enzymatic cycle

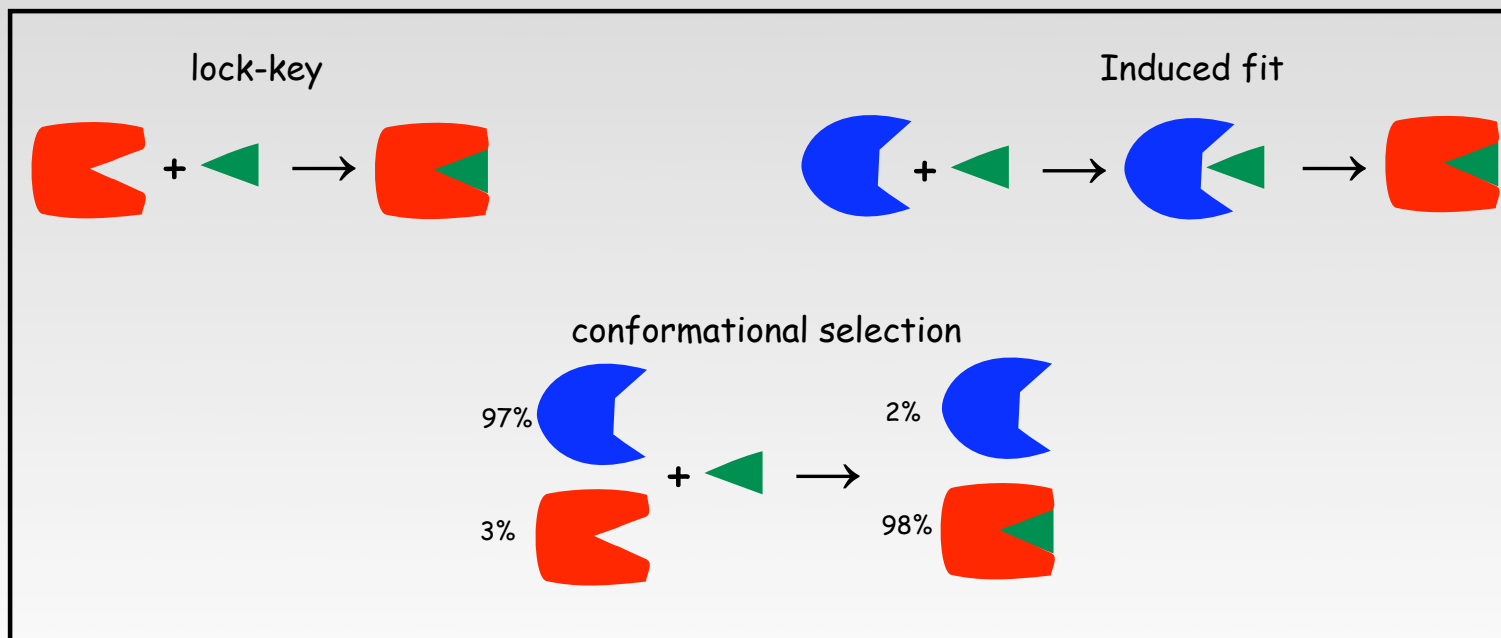
⇒ low populated (high energy) states

⇒ regulation/signalization associated with conformational changes

⇒ interactions

## $\mu$ s-ms motions and proteins function

- ⇒ Proteins : multiple conformers at thermal equilibrium in solution
  - ⇒ XRay cristallography : Selection of one conformation (low energy) in the cristallisation process
  - ⇒ NMR : time averaged conformational ensemble
- ⇒ Ground state > 90%
- ⇒ High energy states, low populated : roles in substrates binding, enzymatic cycles, folding processes (transition states) ...



Boehr et al, Chem. Rev. 2006, 106, 3055-3079

## $\mu$ S-ms motions and proteins function : Practical course

- ⇒ Analysis of the relaxation dispersion pulse sequence
  - ⇒ The relaxation compensated CPMG scheme
  - ⇒ Experimental protocol

- ⇒ The understanding of molecular mechanisms underlying proteins functions necessitates the characterization of the dynamics of the macromolecular systems, and of the minor transient states along the functional pathways.
- ⇒ Some recently published studies based on the use of relaxation dispersion experiments to investigate these questions will be discussed.